



TEST

Practice Test (7C: Amino acids)





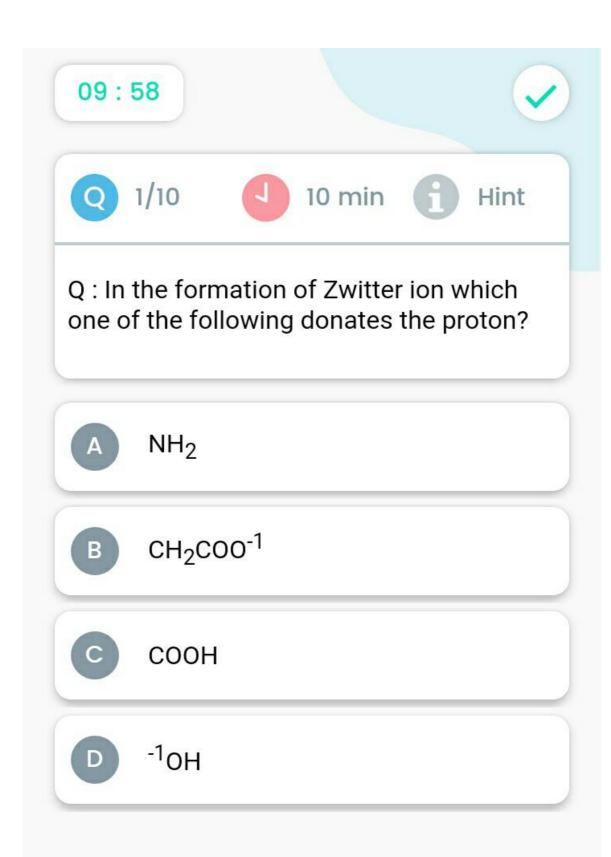


10 min

Topics

General Structure of Amino Acids found in Proteins, Zwitter Ion, Acid base properties of Amino Acids., Amino Acids on the basis of Nature of Rgroup., Peptide bond formation

Start Test













Q: All the naturally occurring amino acids are

- α amino acids
- β amino acids
- γ amino acids
- δ amino acids



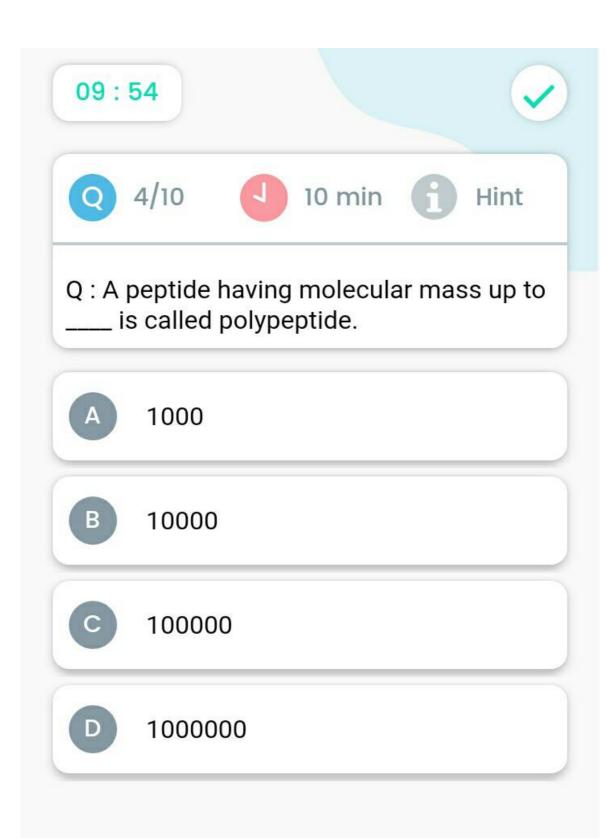


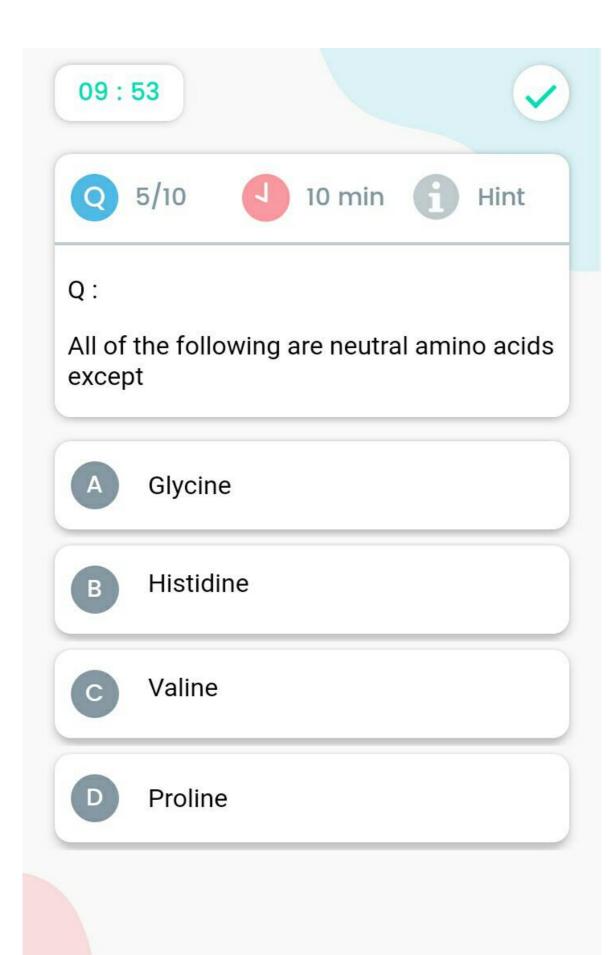
- 3/10
- 10 min 👔

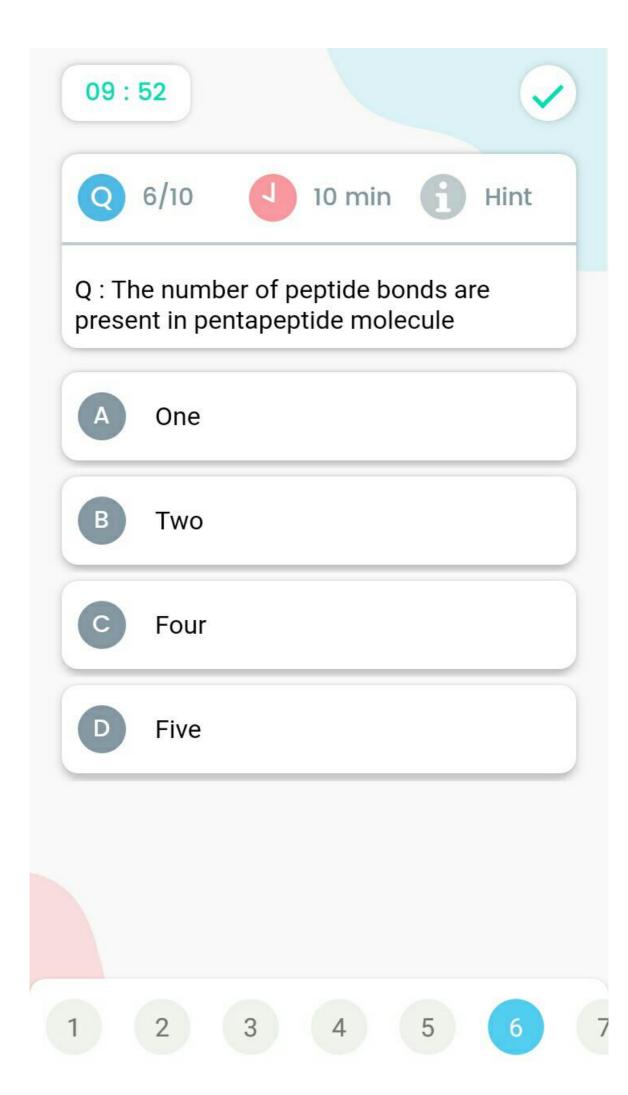
Hint

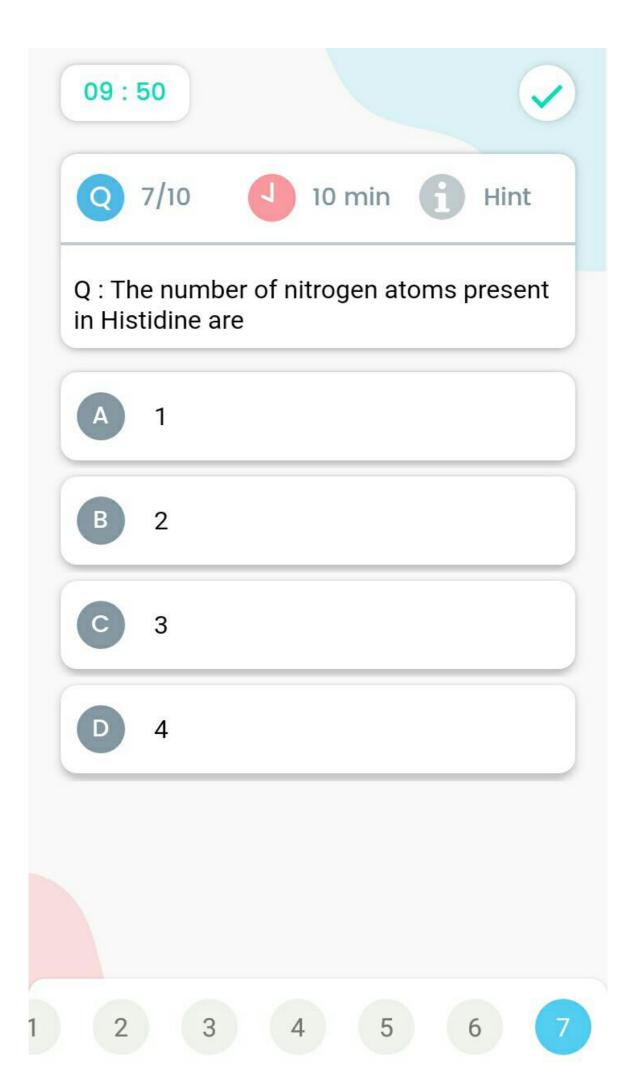
Q : Peptide linkage is similar to

- Ester linkage
- ether linkage
- amide linkage
- none of these













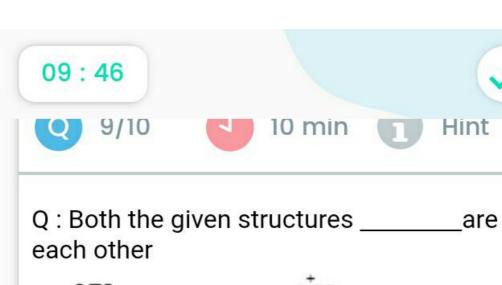


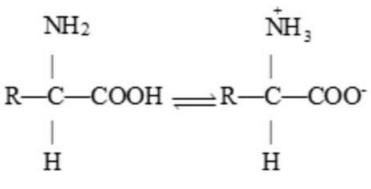




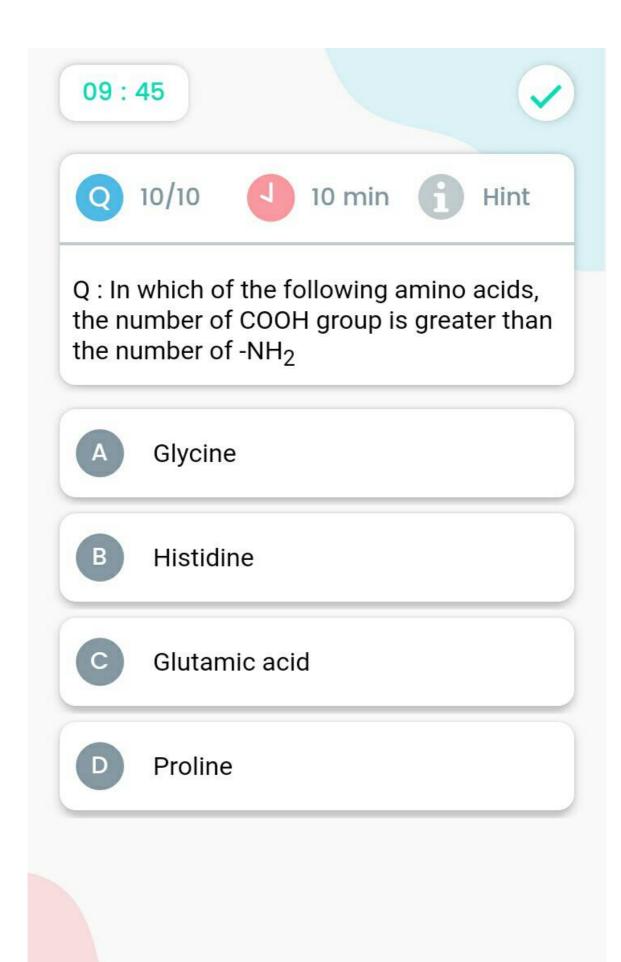
Q: Number of amine group present in lysine and valine are

- 2,1
- 1,2
- 2,2
- 1,1

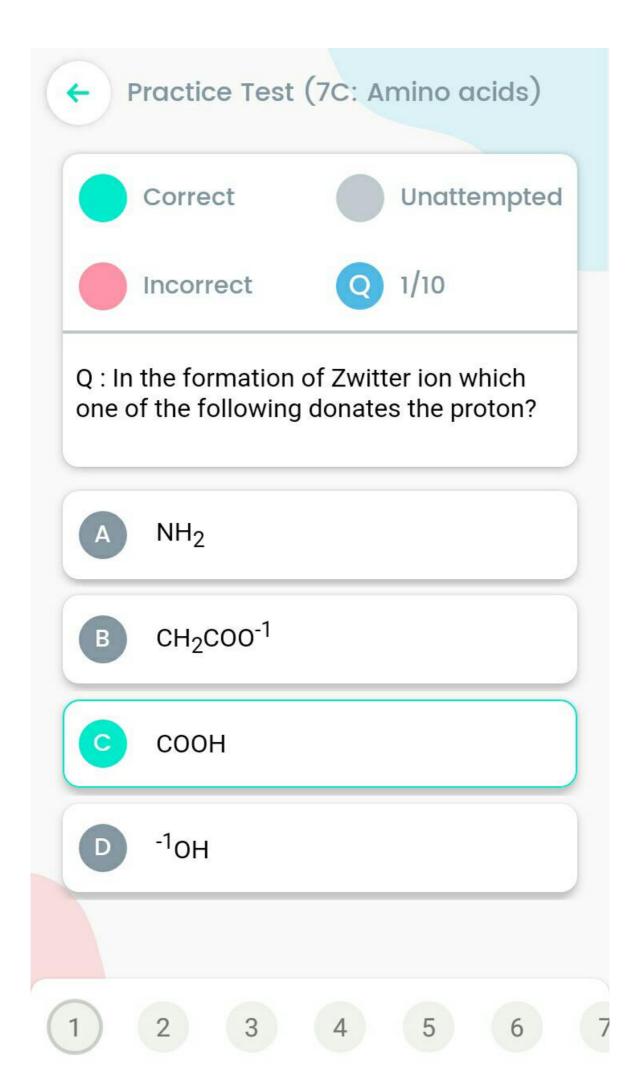


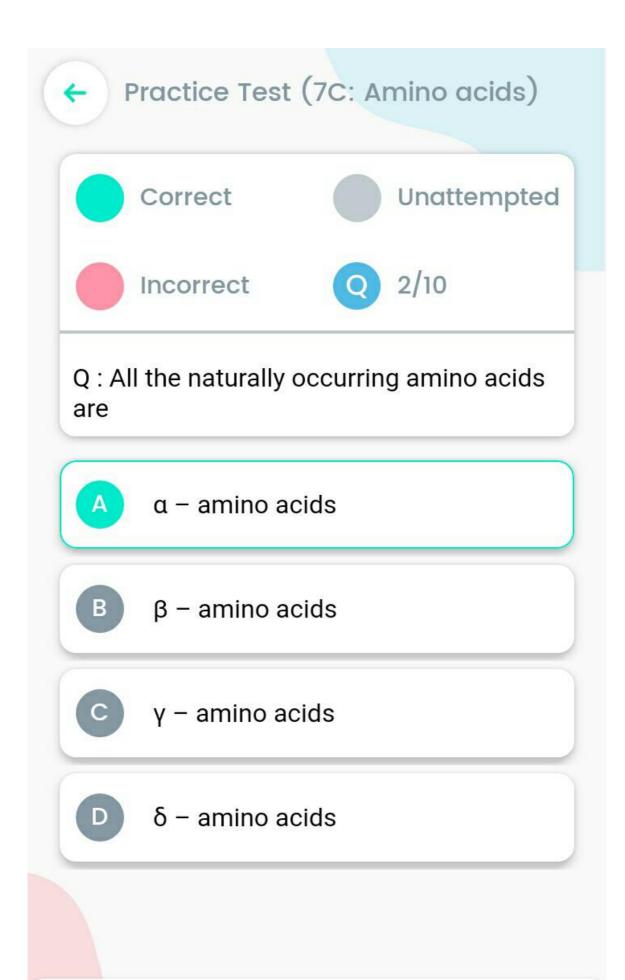


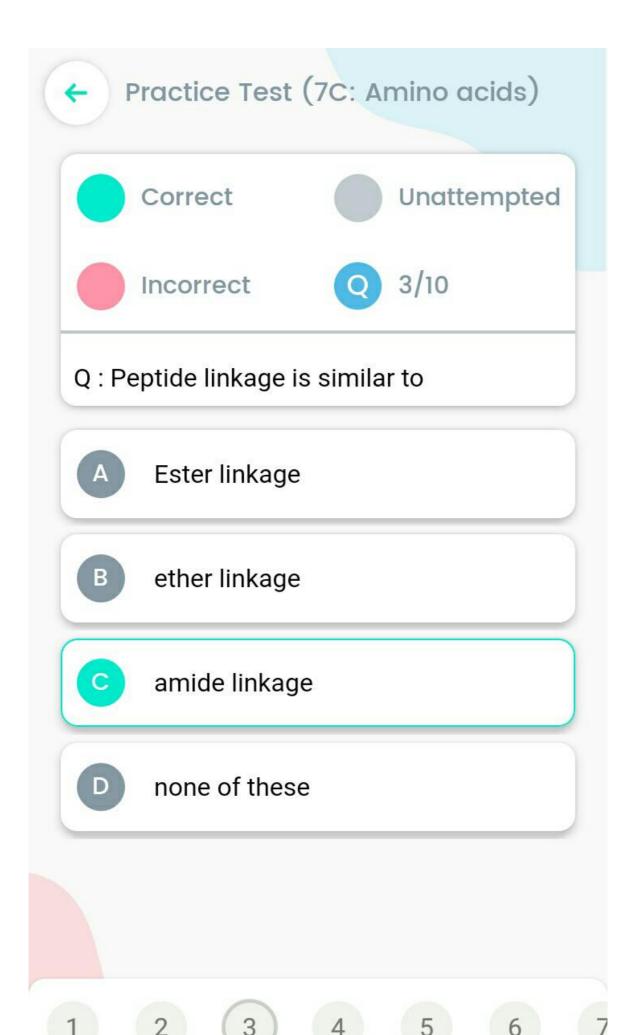
- A Metamers
- B Tautomers
- C Isotopes
- Functional group isomers

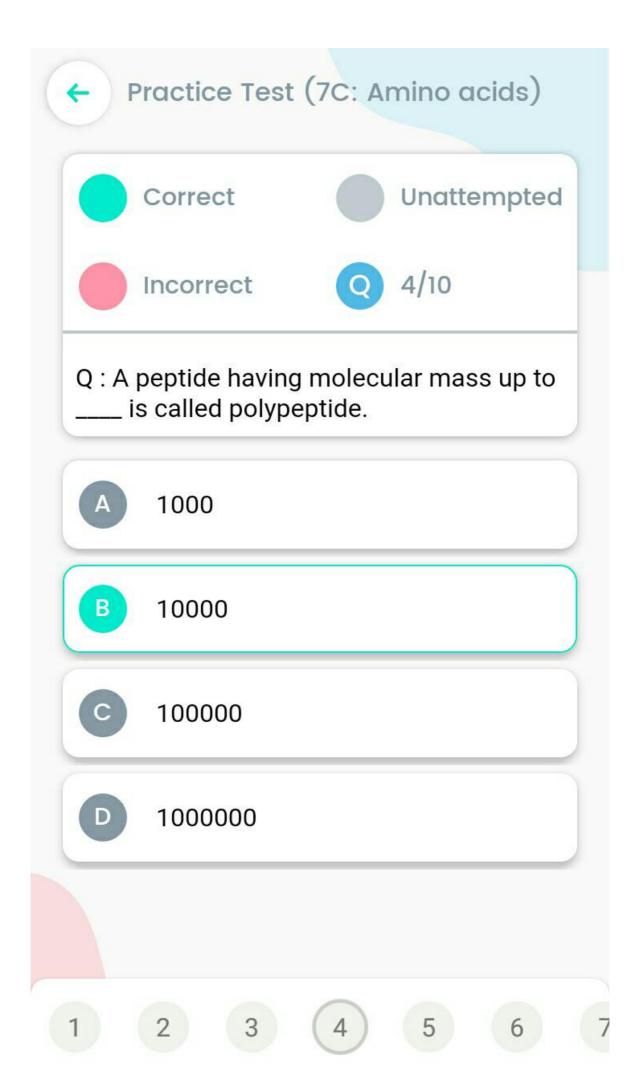


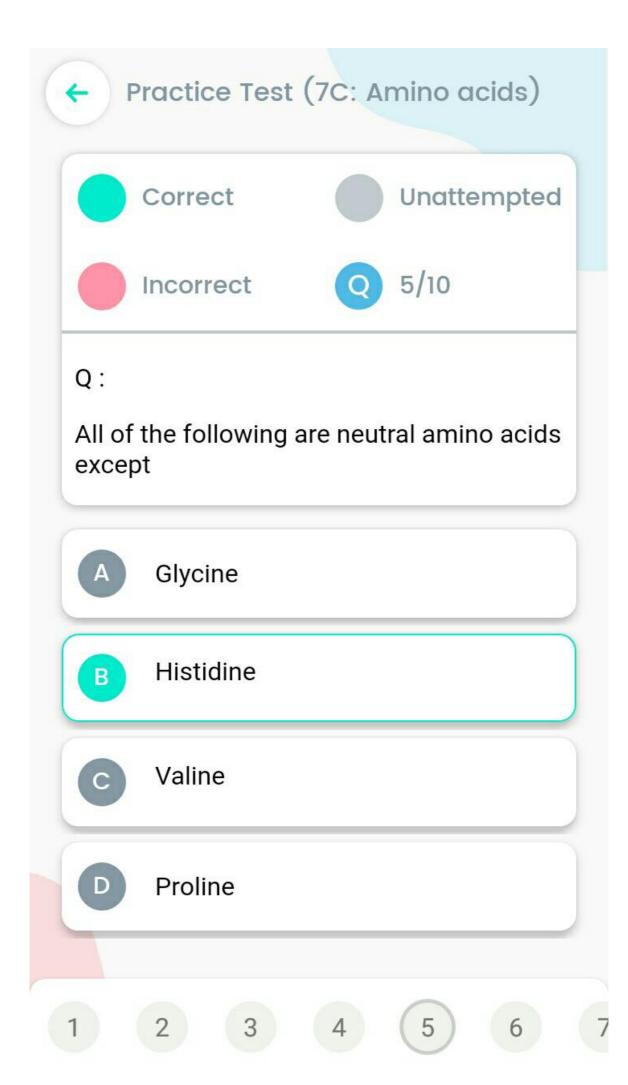
4 5 6 7 8 9 10

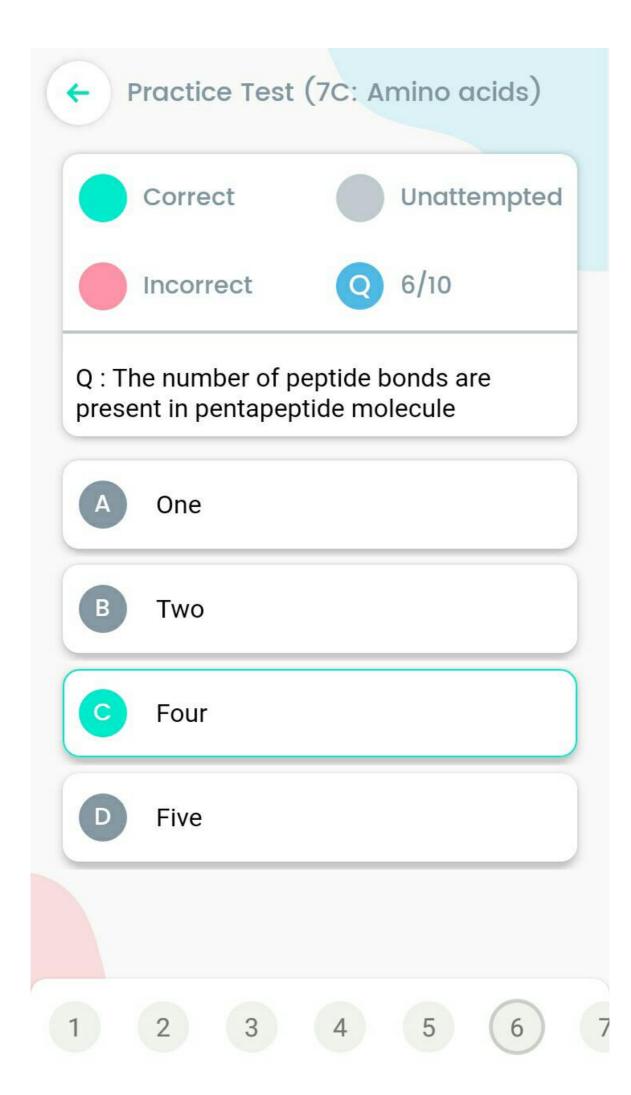


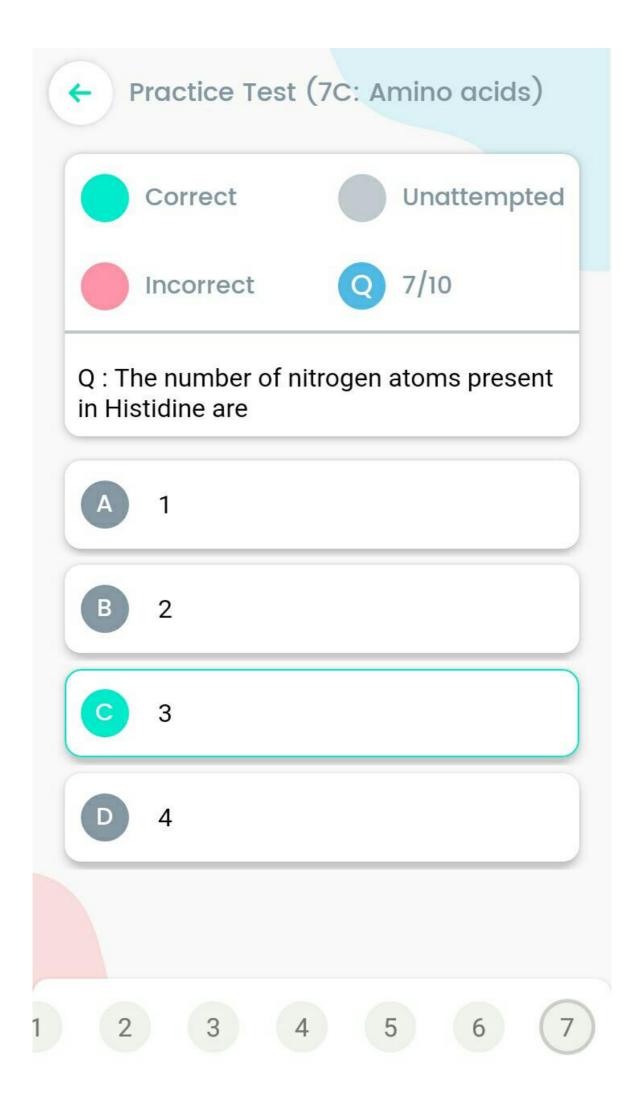


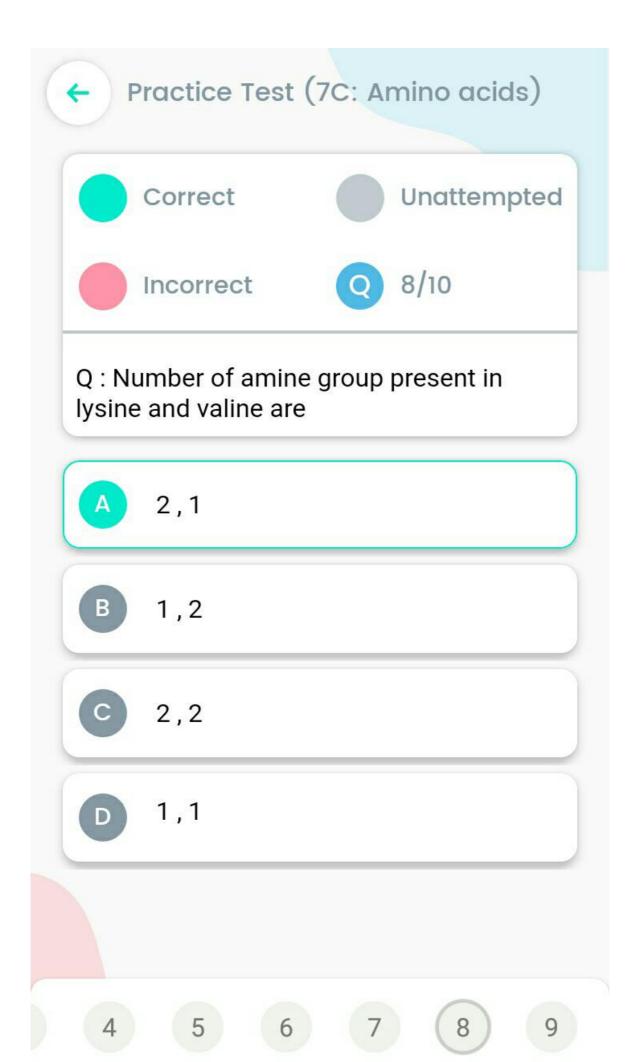










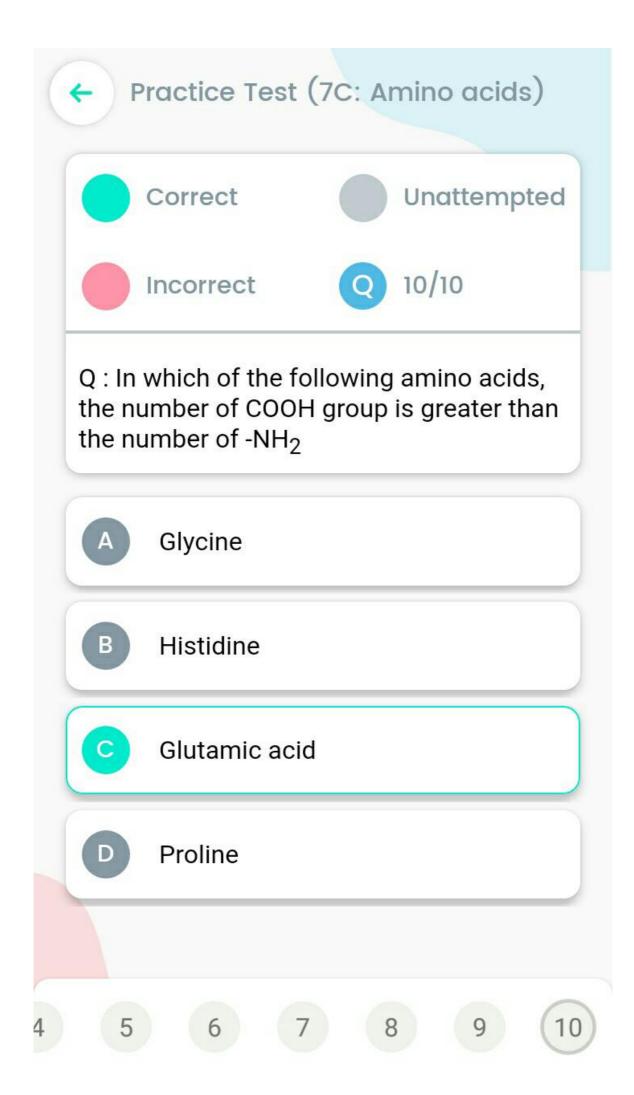






Q : Both the given structures _____are each other

- A Metamers
- B Tautomers
- C Isotopes
- Functional group isomers







TEST

Test Level-1 (7C: Amino acids)





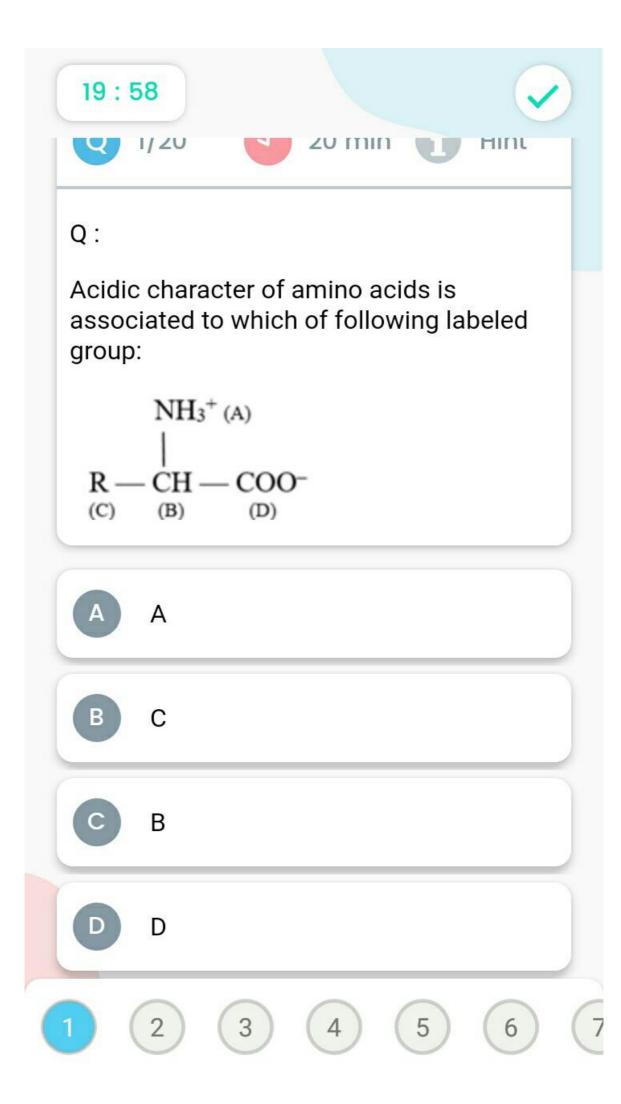


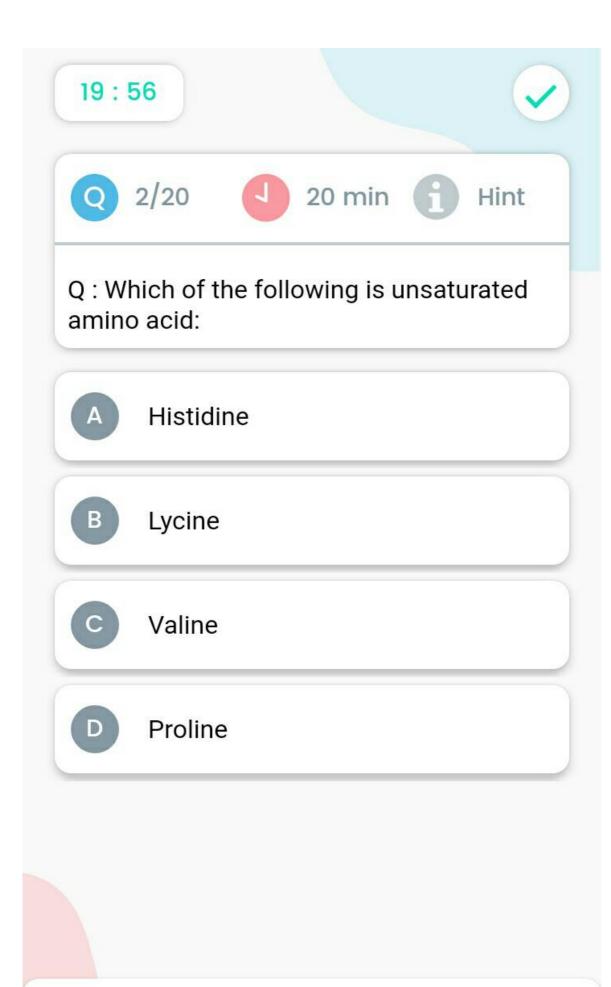
20 min

Topics

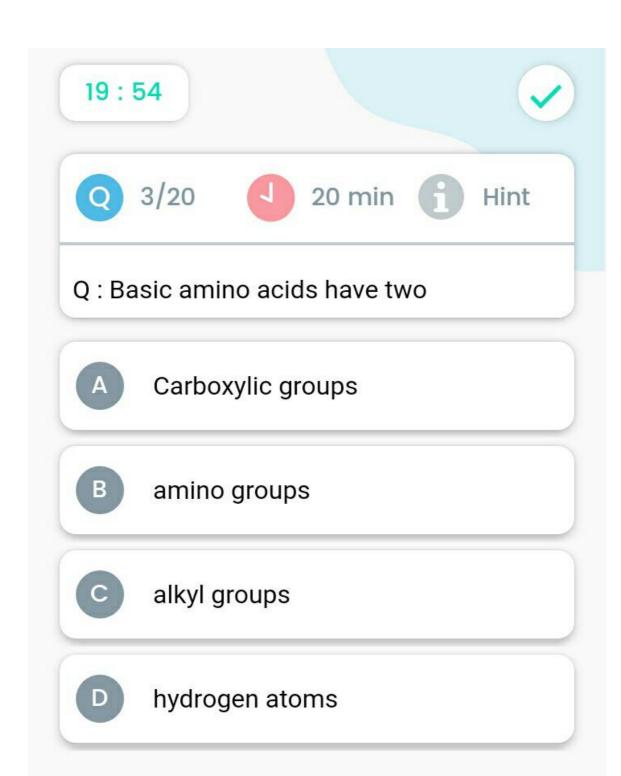
General Structure of Amino Acids found in Proteins, Zwitter Ion, Acid base properties of Amino Acids., Amino Acids on the basis of Nature of Rgroup., Peptide bond formation

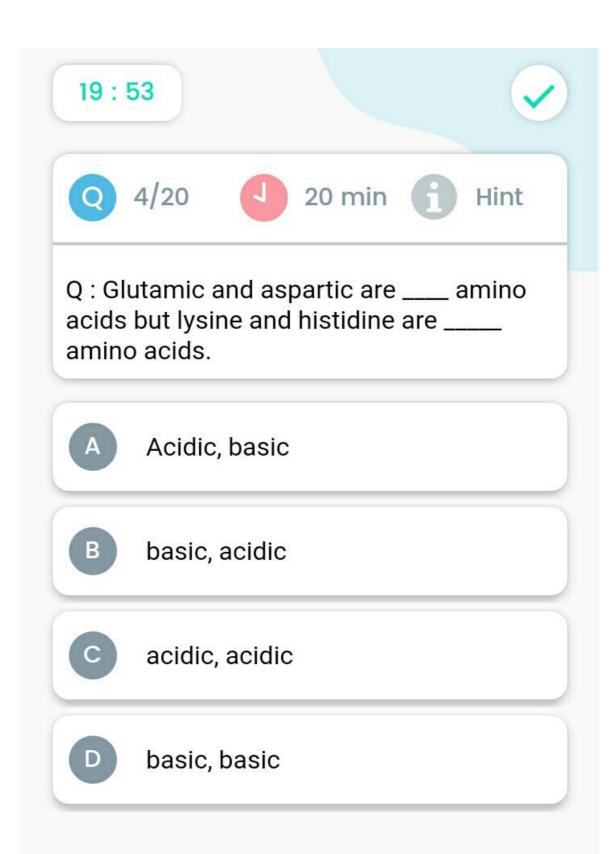
Start Test



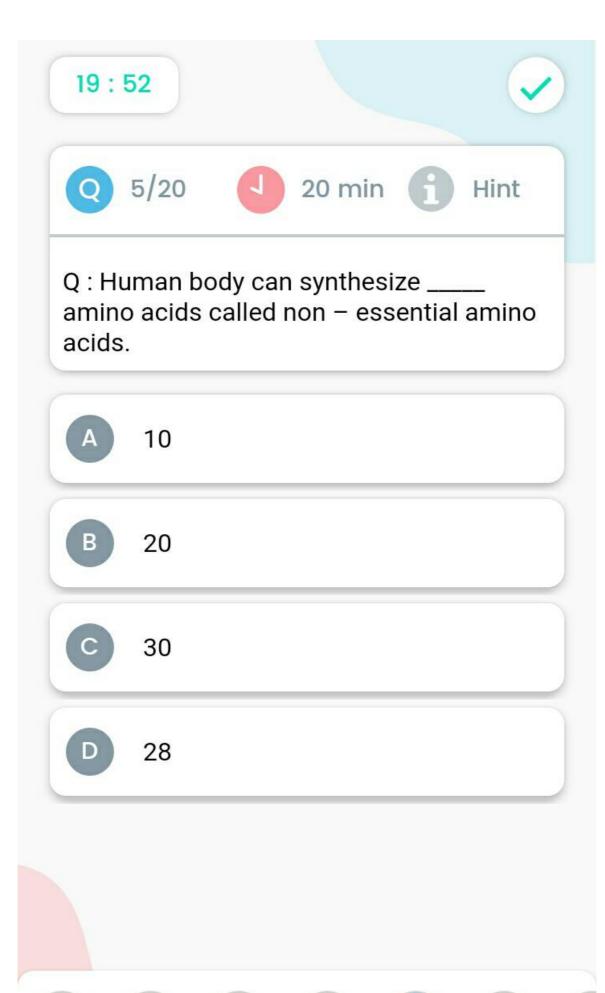


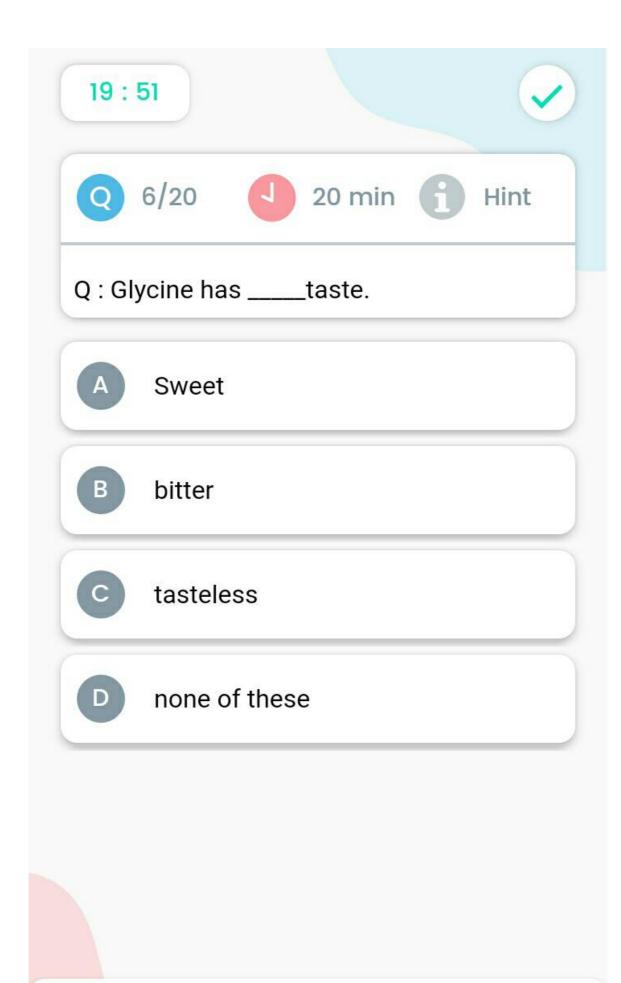


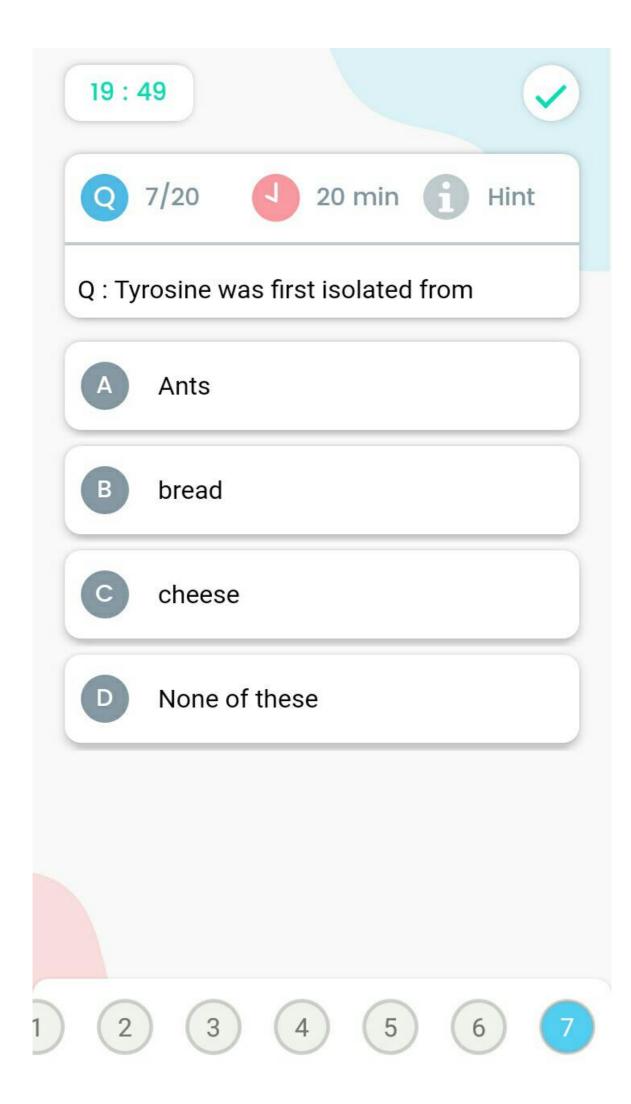


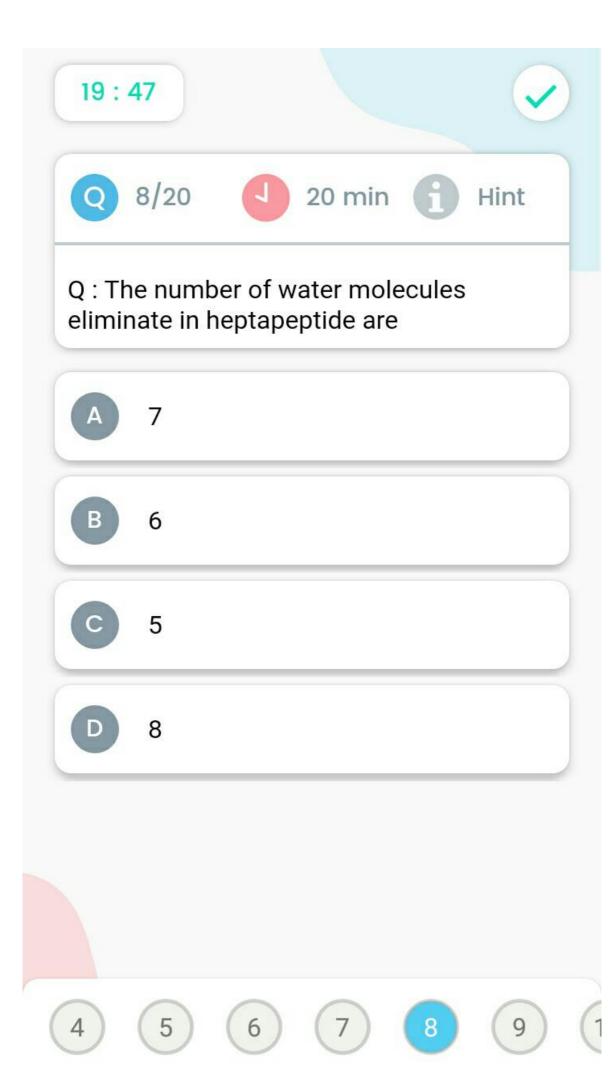


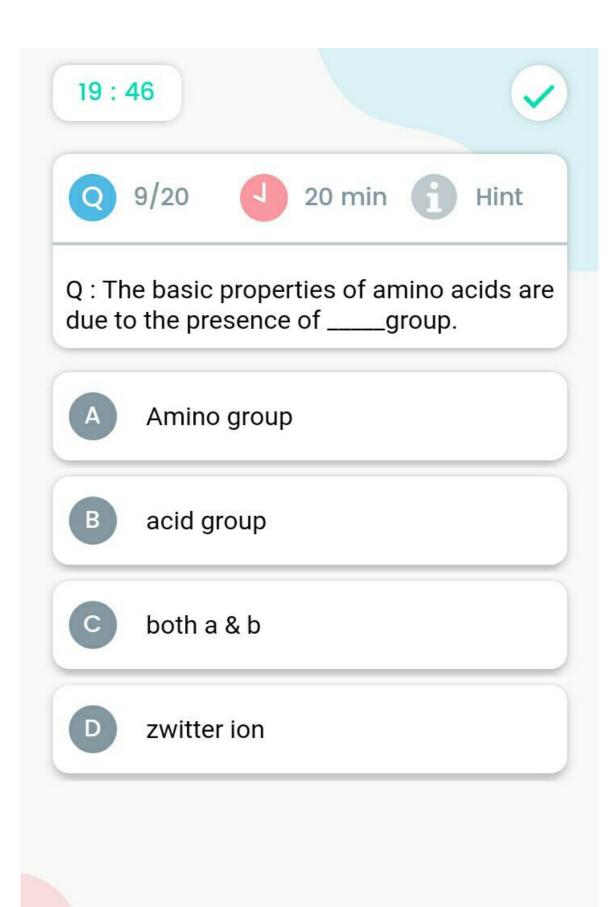














19:44



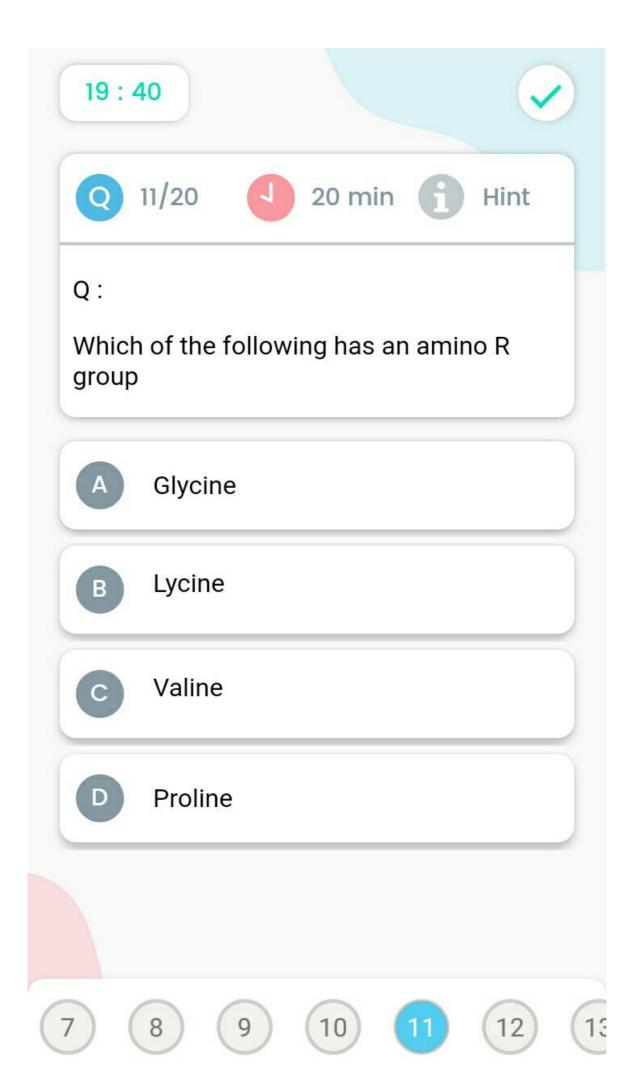
Q: The group which is different in glycine and alanine?

- A A
- B B
- C C
- D D
- 4
- 5
- 6
- 7
- 8
- 9
- 10



Q : The group which is different in glycine and alanine?

- A A
- B B
- CCC
- D











20 min



Hint

Q: Which of the following structures is not an α -amino acid?

- NH₂-CH(CH₃)-COOH
- B (CH₃)₂CHCH(NH₂)COOH
- NH₂-CH₂-CH₂-CH₂COOH
- NH₂CHCOOH | CH₂C₆H₅

7

8

9

10

11

12

13







13/20



20 min



Hint

Q : In formation of protein, carboxylic group of one amino acid and amino group of other amino acid condensed together to give

- A Peptide linkage
- B Ester linkage
- 1 → 6 glycosidic linkage
- β 1-4 Glycosidic linkage.

7

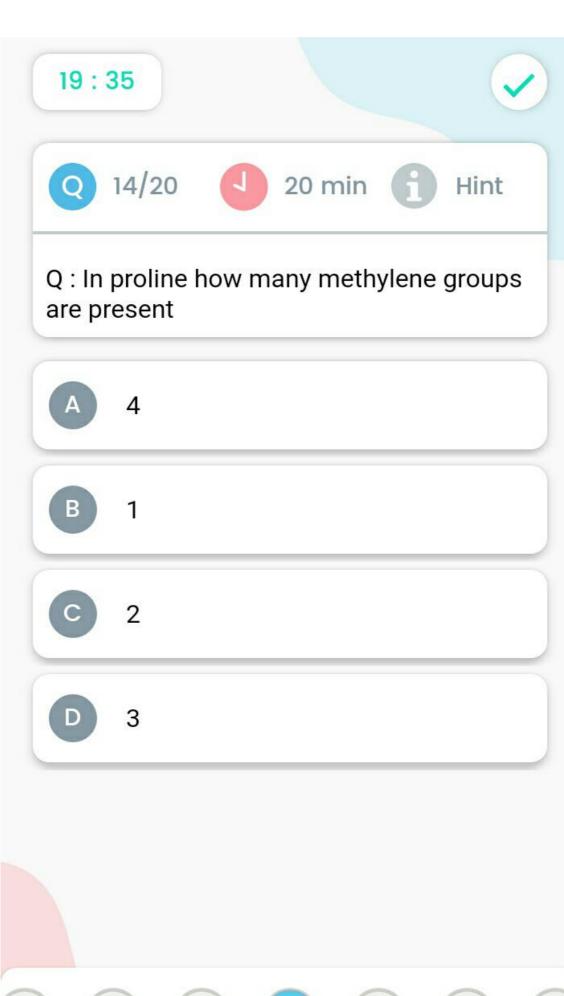


9

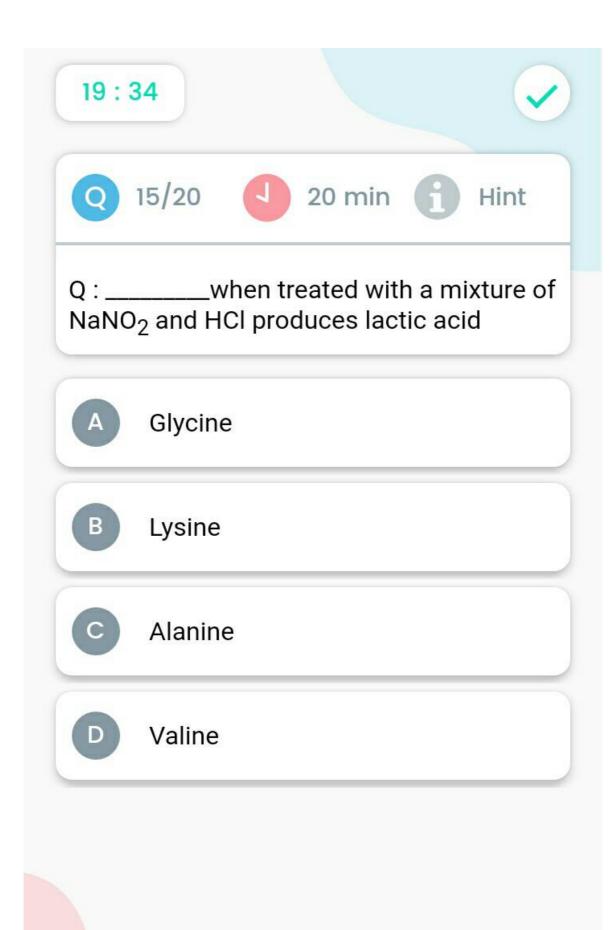








11 (12) (13) (14) (15) (16) (17)



11 12 13 14 15 16 17









20 min



Hint

Q : A polypeptide is conventionally called a protein if it has

- Less than 10,000 amino acid units
- Molar mass more than 10,000 (g/mol)
- More than 10,000 amino acid units
- Molar mass less than 10,000 (g/mol)

11

12

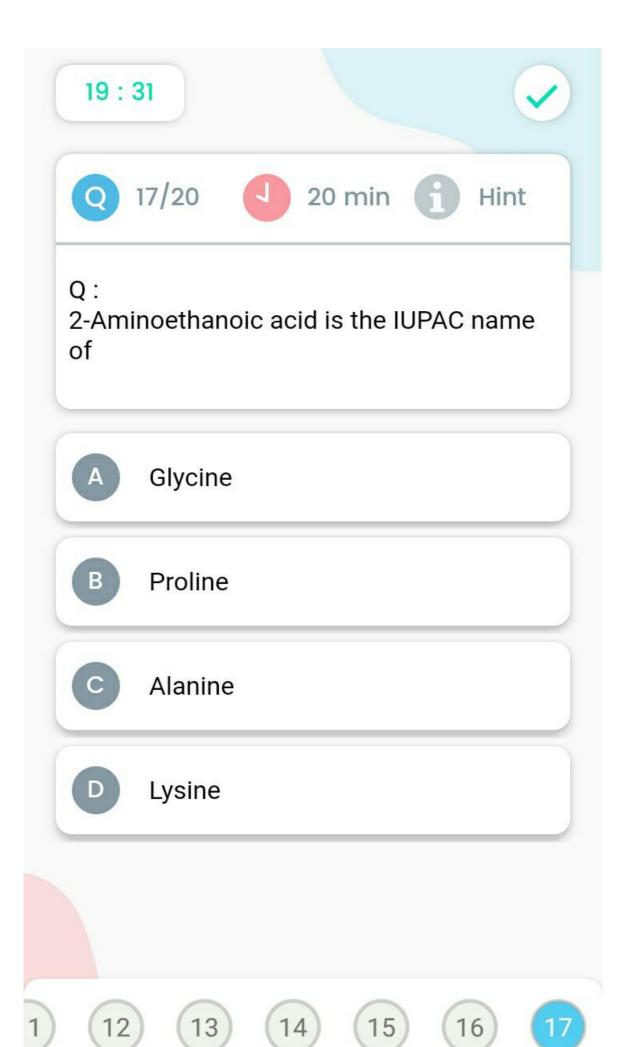
13

14

15

16

17









18/20



20 min 🚹 Hint



Q: Number of amine group present in lysine and glycine are



2,1



1,2



2,2



1,1









20 min



Hint

Q: Which one of the following reagent is used for identification of amino acid?

- A Fehling's solution
- B Ninhydrin
- Bendict's solution
- Copper sulphate

14

15

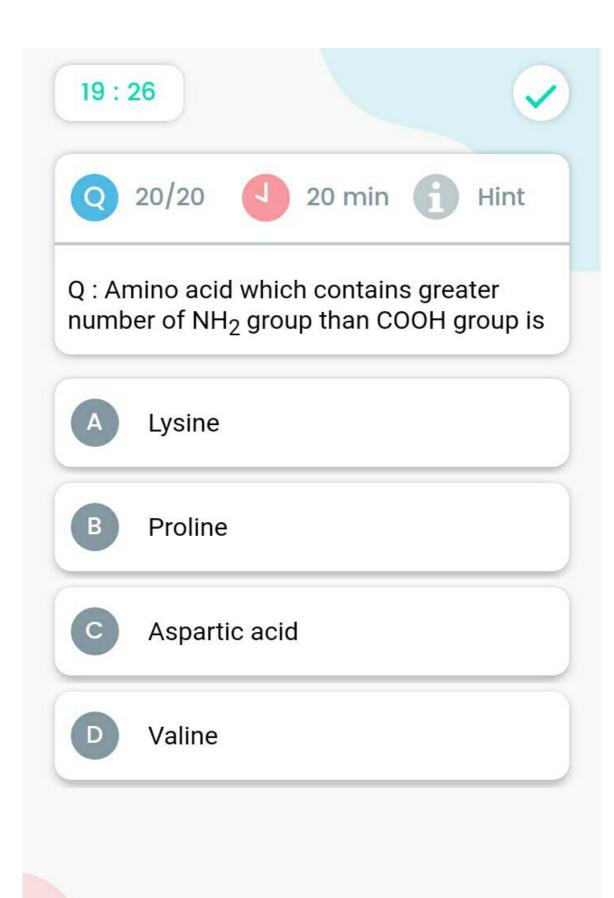
16

17

18

19

20

















Incorrect

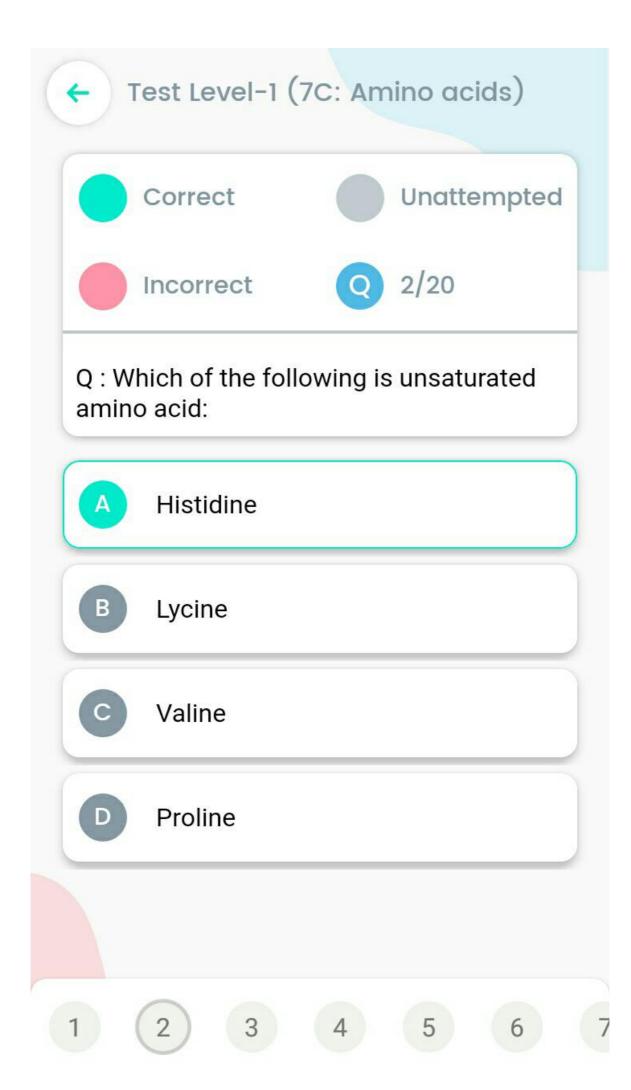


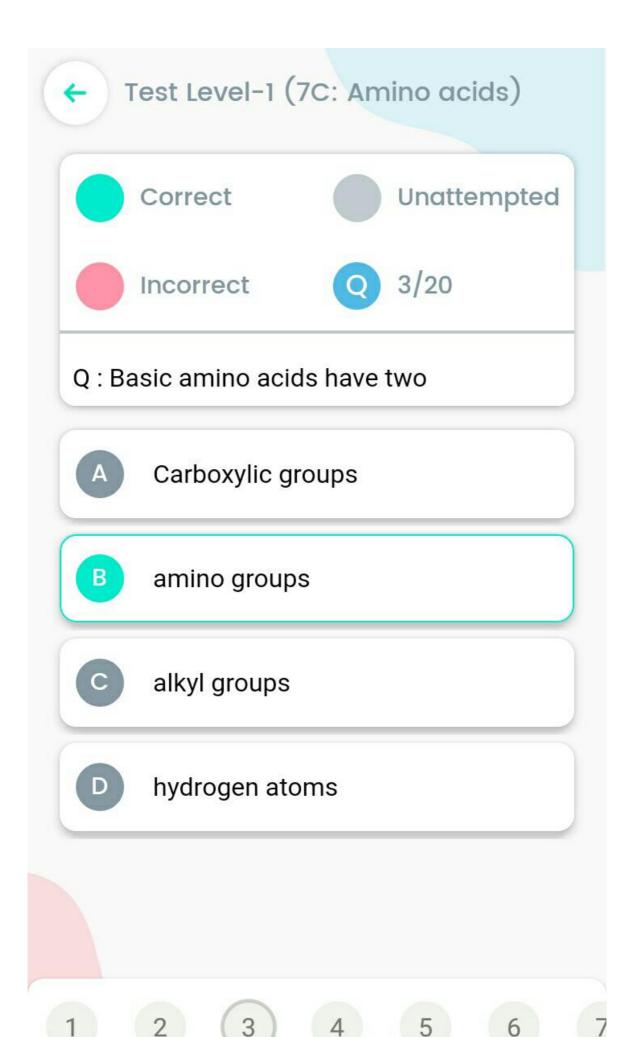
1/20

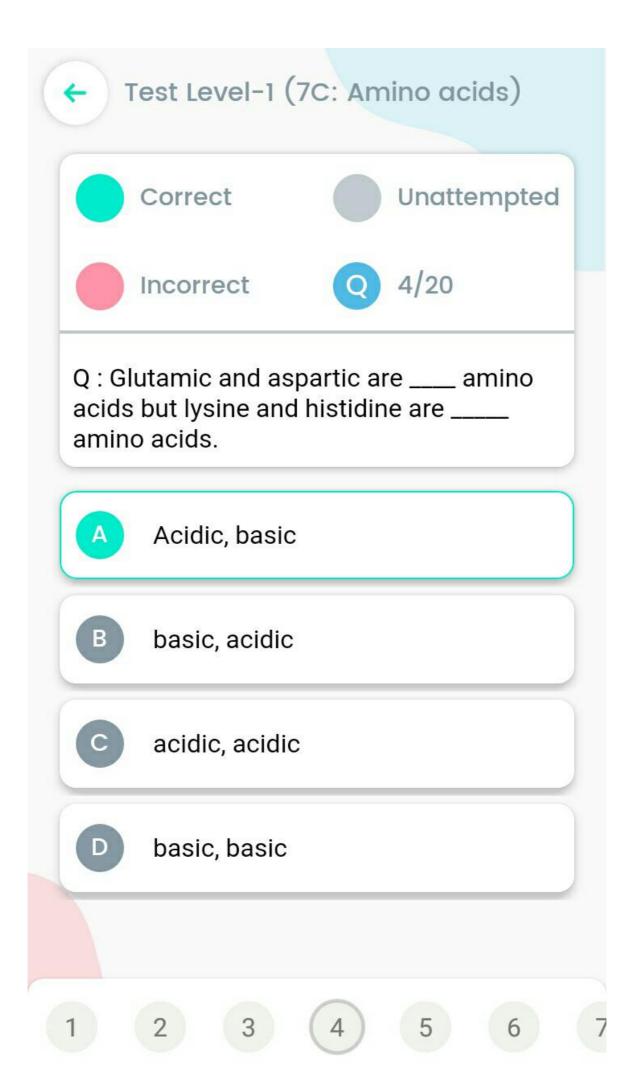
Q:

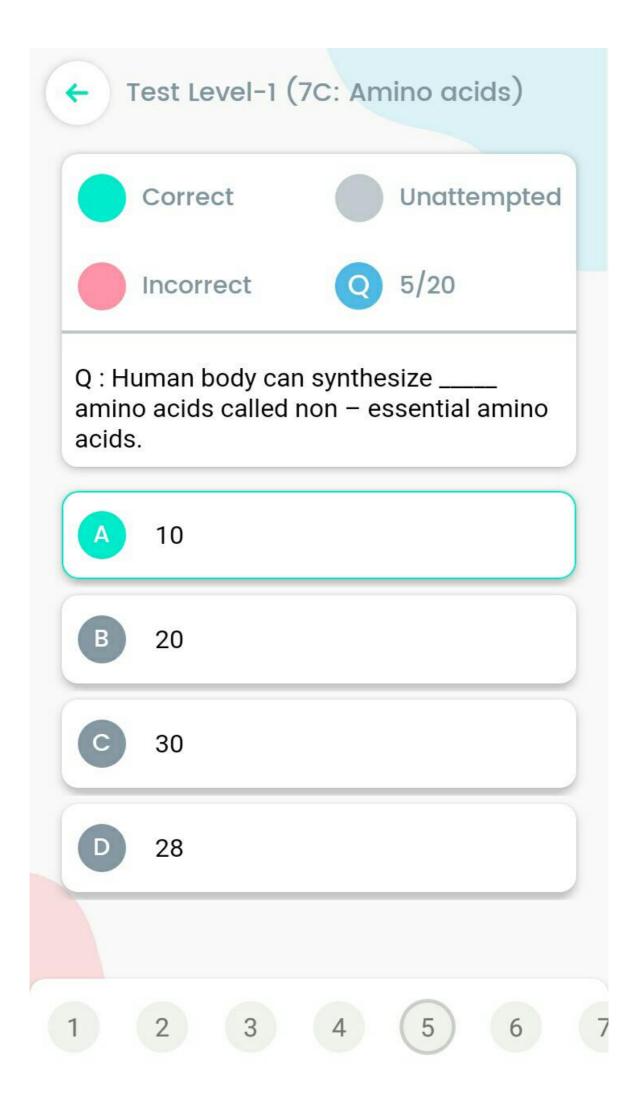
Acidic character of amino acids is associated to which of following labeled group:

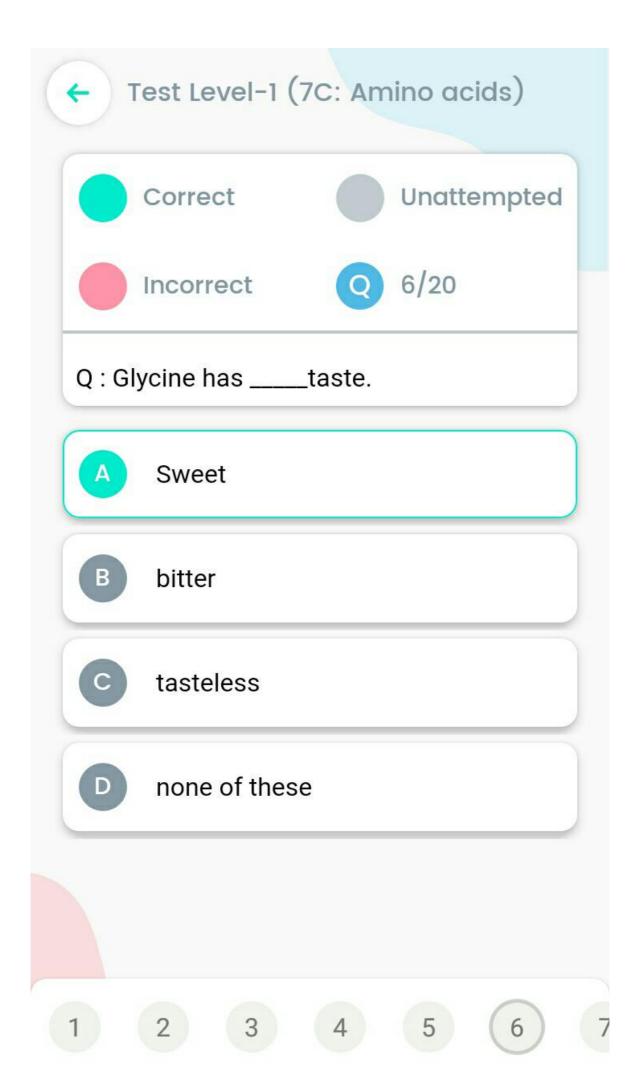
- A A
- B C
- C B
- D D

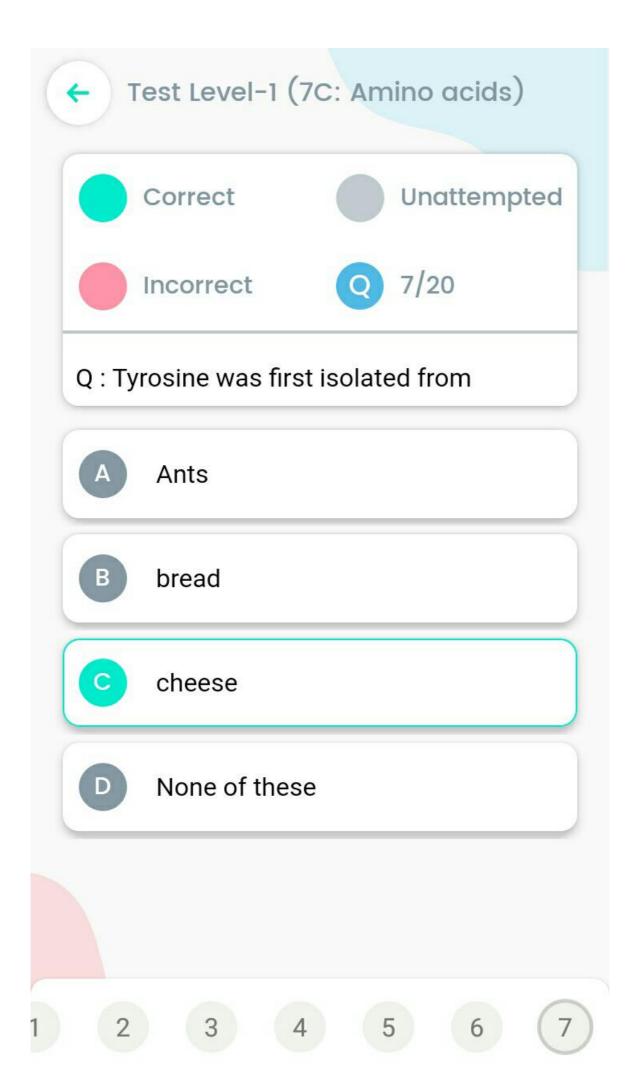


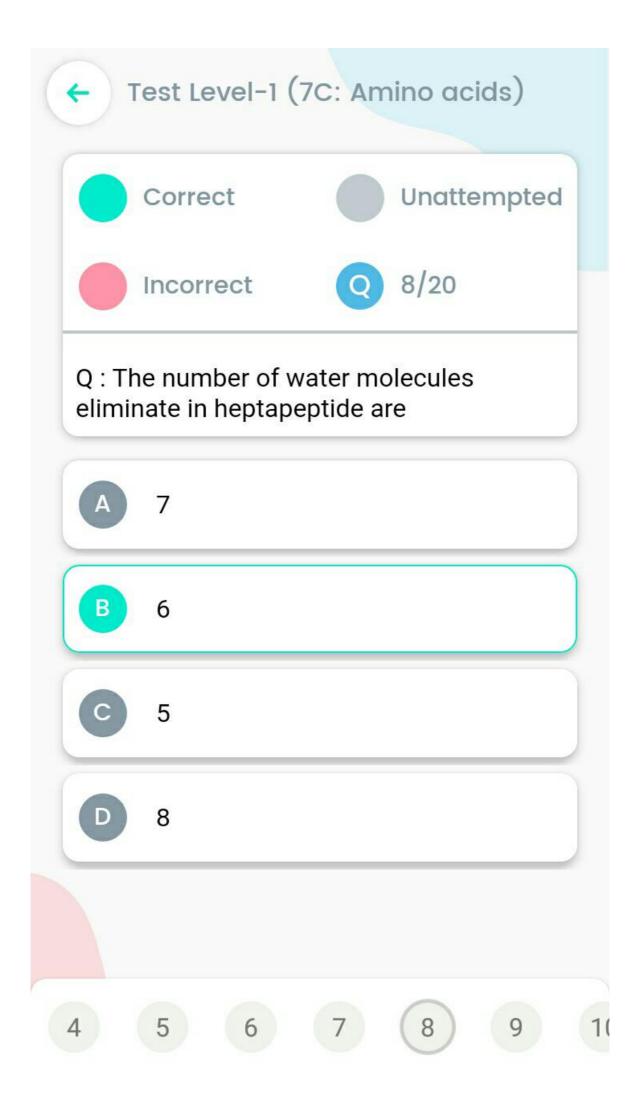


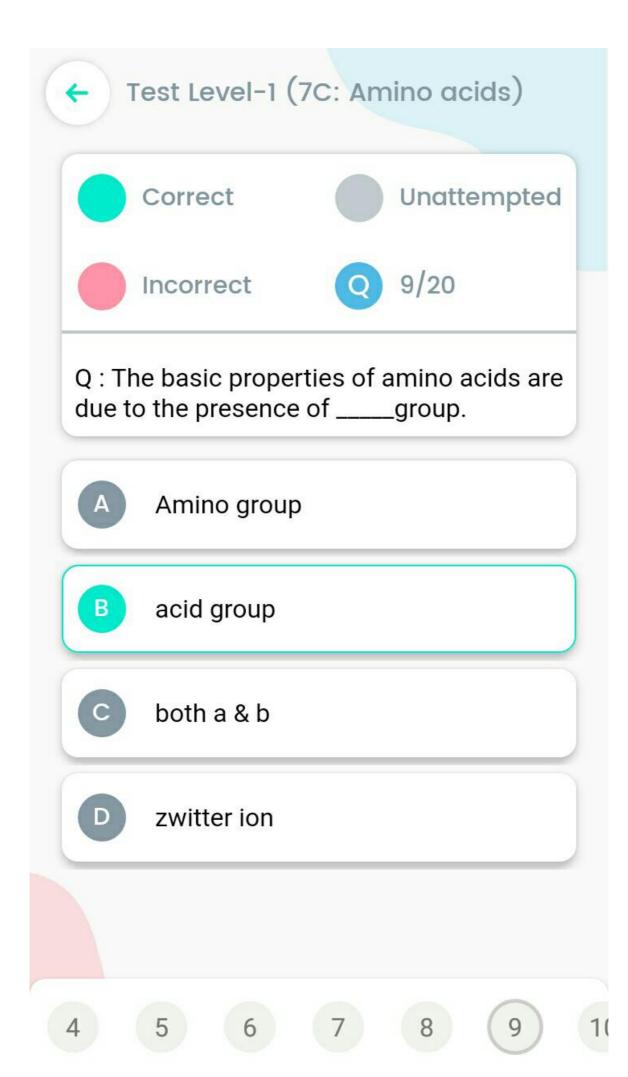
















Q: The group which is different in glycine and alanine?

AA



Test Level-1 (7C: Amino acids)



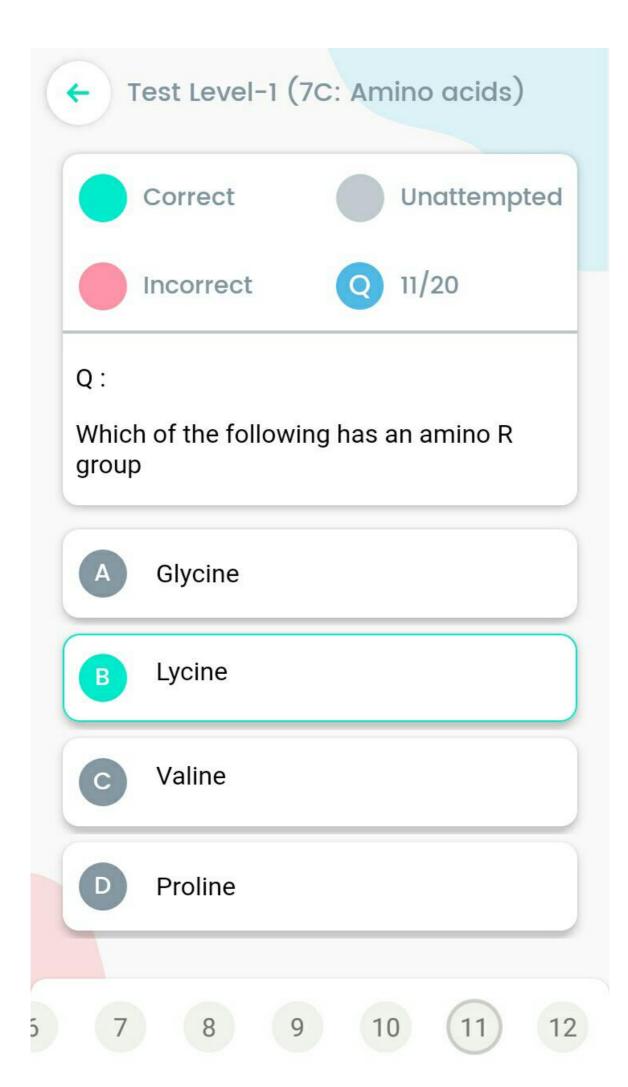
.........

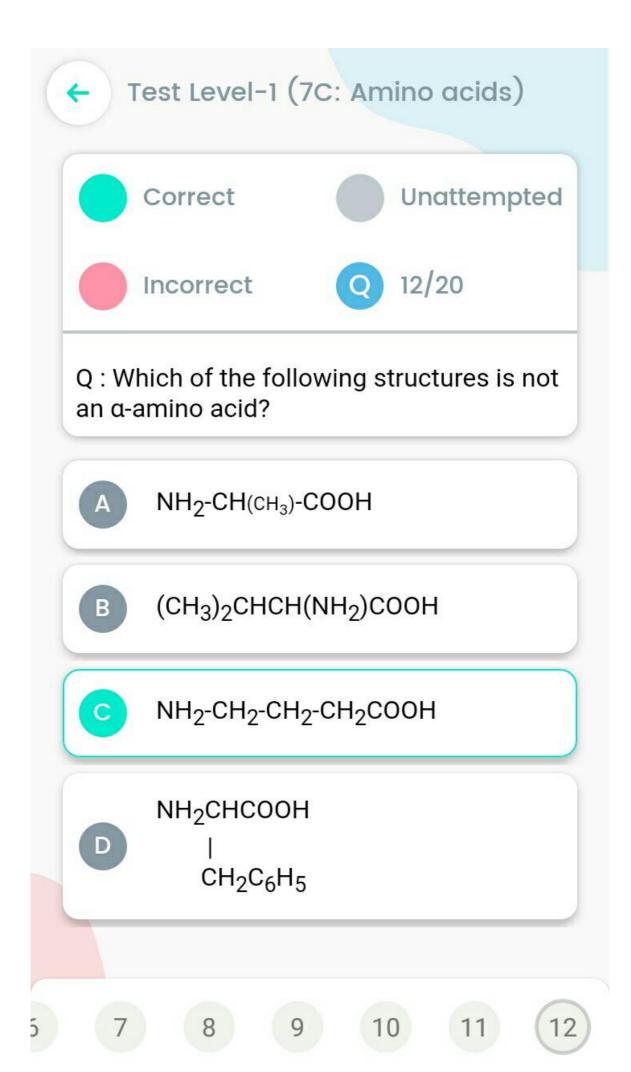


. ~ / ~ ~

Q: The group which is different in glycine and alanine?

- A A
- ВВ
- C C





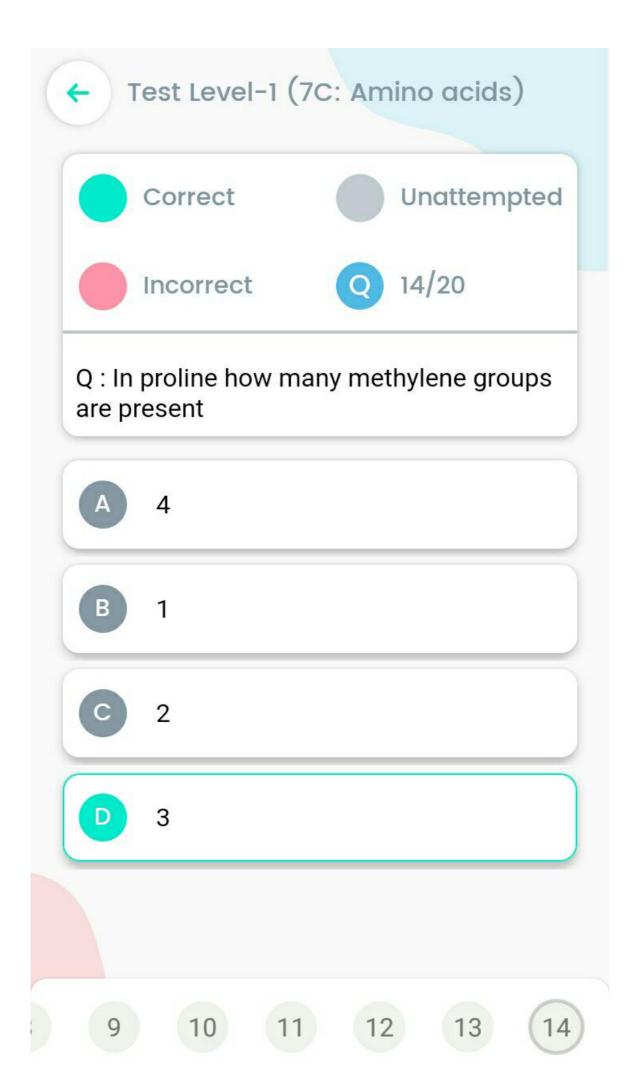


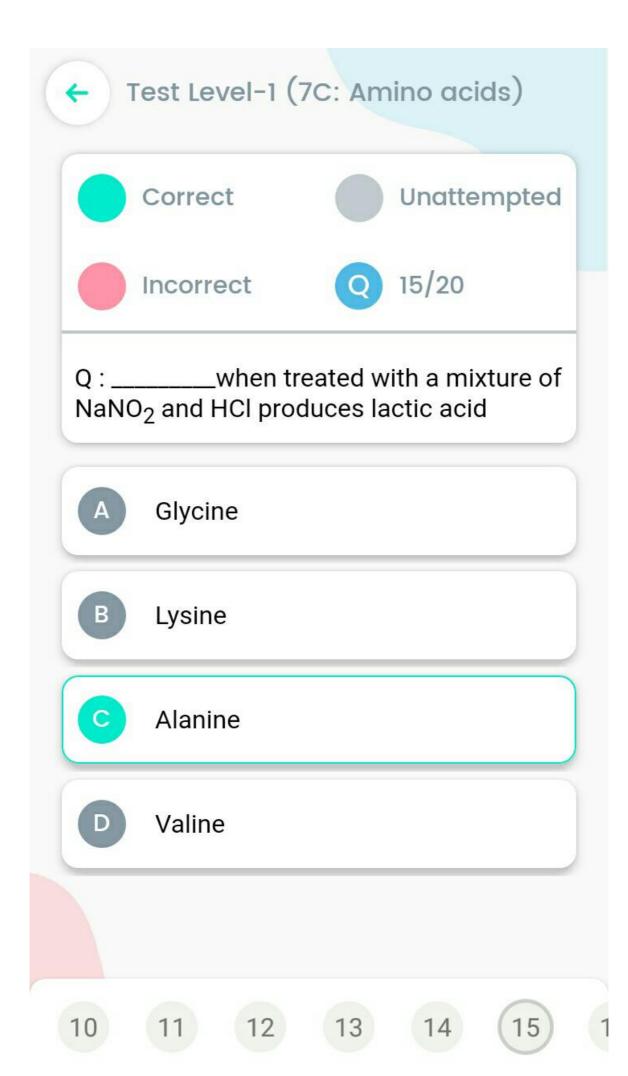




Q : In formation of protein, carboxylic group of one amino acid and amino group of other amino acid condensed together to give

- A Peptide linkage
- B Ester linkage
- 1 → 6 glycosidic linkage
- β 1-4 Glycosidic linkage.





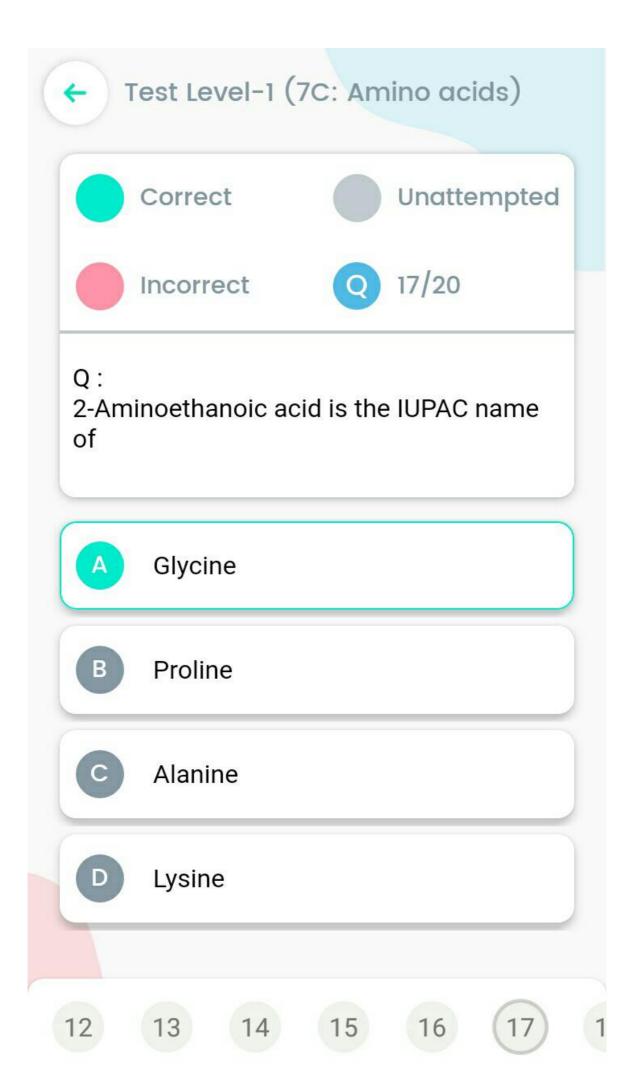


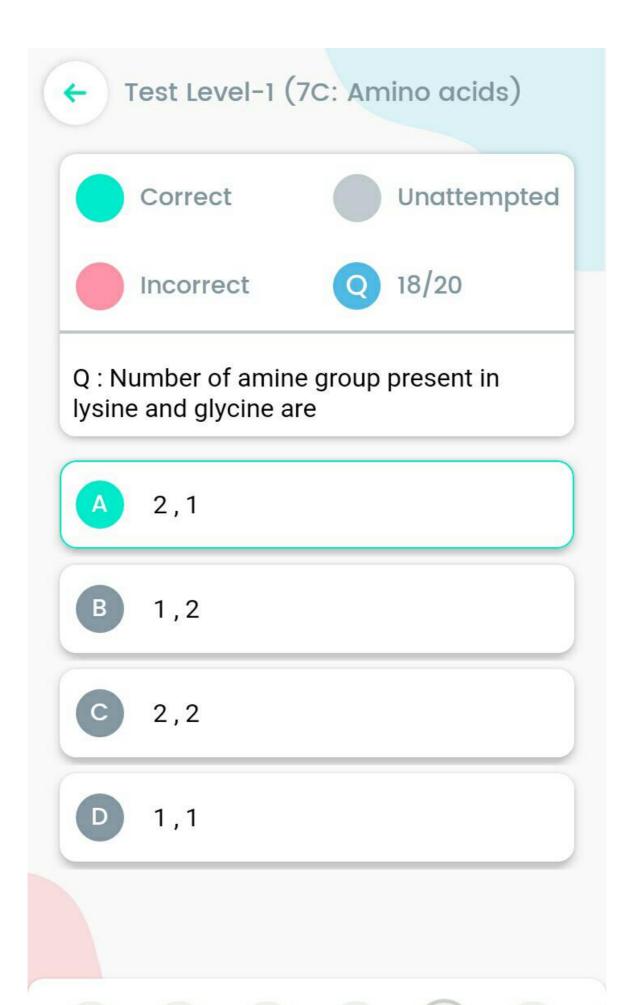


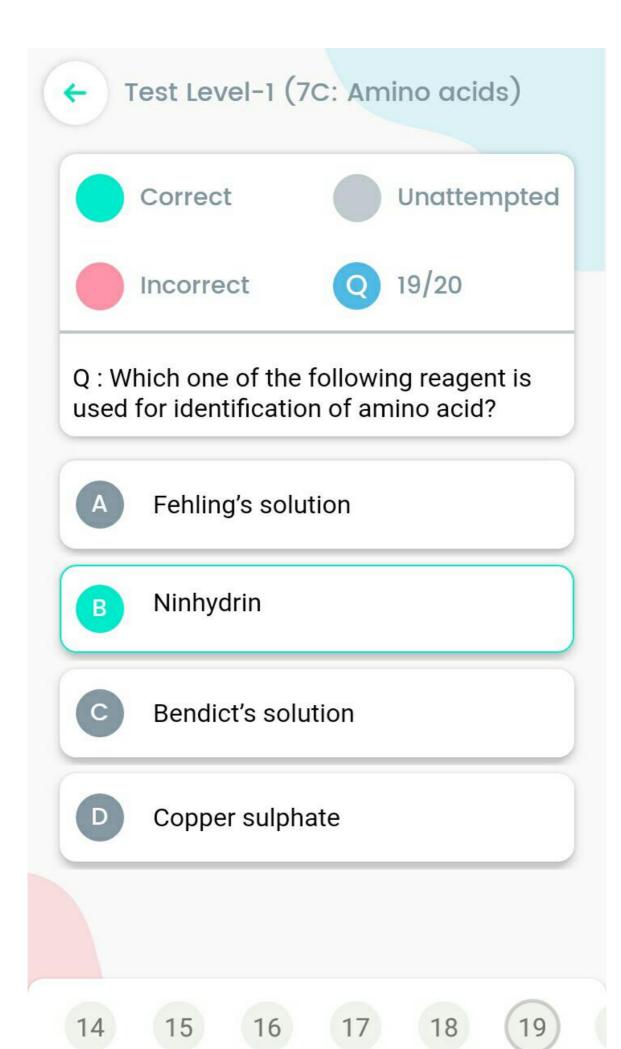
Q : A polypeptide is conventionally called a protein if it has

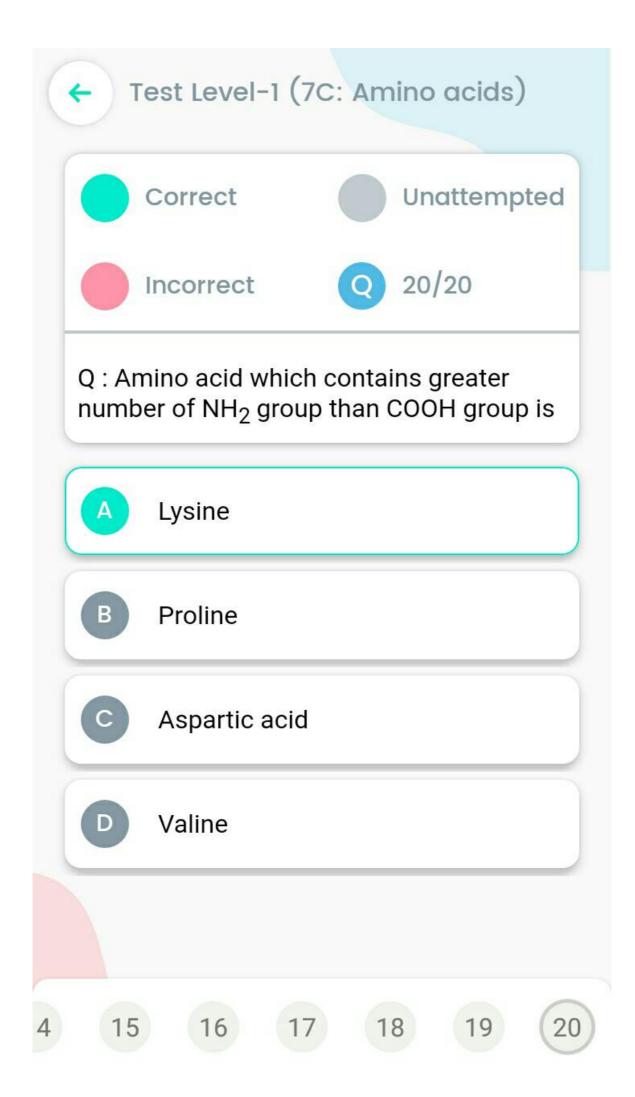
- Less than 10,000 amino acid units
- Molar mass more than 10,000 (g/mol)
- More than 10,000 amino acid units
- Molar mass less than 10,000 (g/mol)

12 13 14 15 (16) 17 1













TEST

Test Level-2 (7C: Amino acids)





30 Questions 30 min

Topics

Amino Acids

Start Test









30 min 👔



Hint

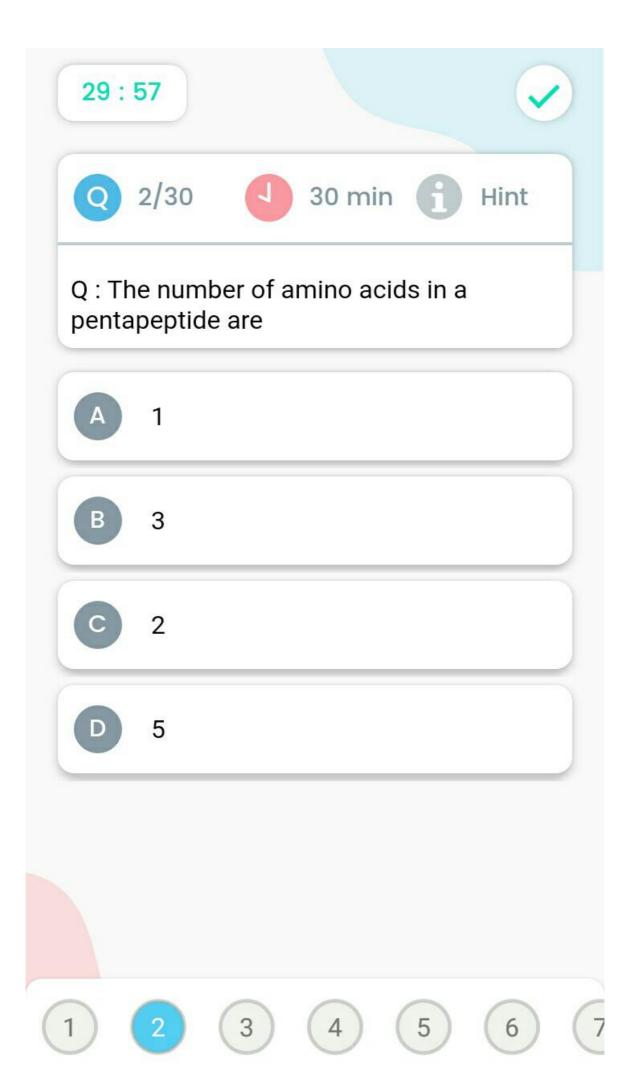
Q: The IUPAC name of alanine is

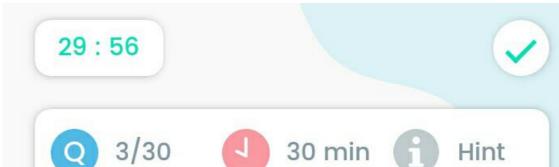
- 2-Aminobutanoic acid
- 2-Aminopropanoic acid
- 2-Ethanoic acid
- 2-Aminopropionic acid







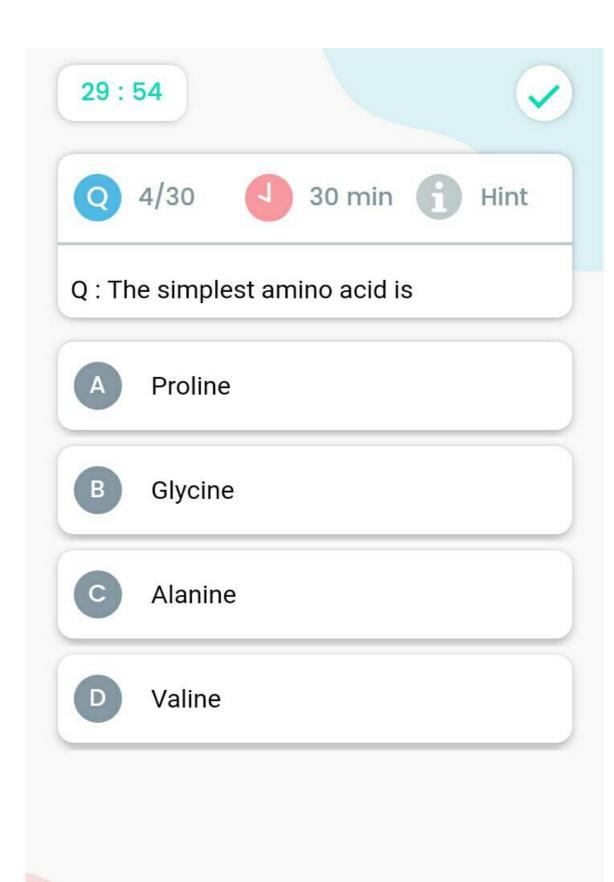




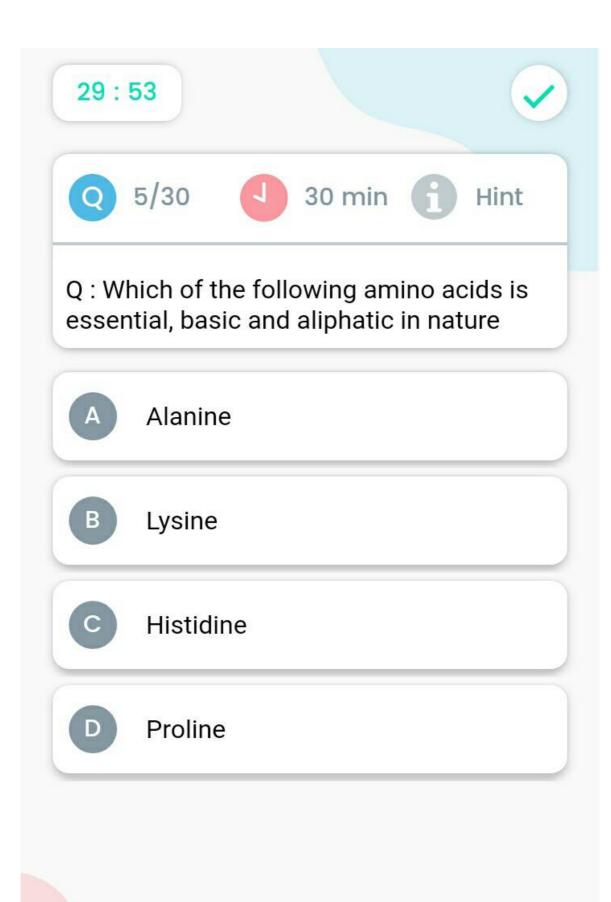
Q : The pair of amino acids that has more number of carboxyl groups as compared to amino groups

- A Histidine and lysine
- B Proline and glycine
- Histidine and alanine
- Aspartic acid and glutamic acid

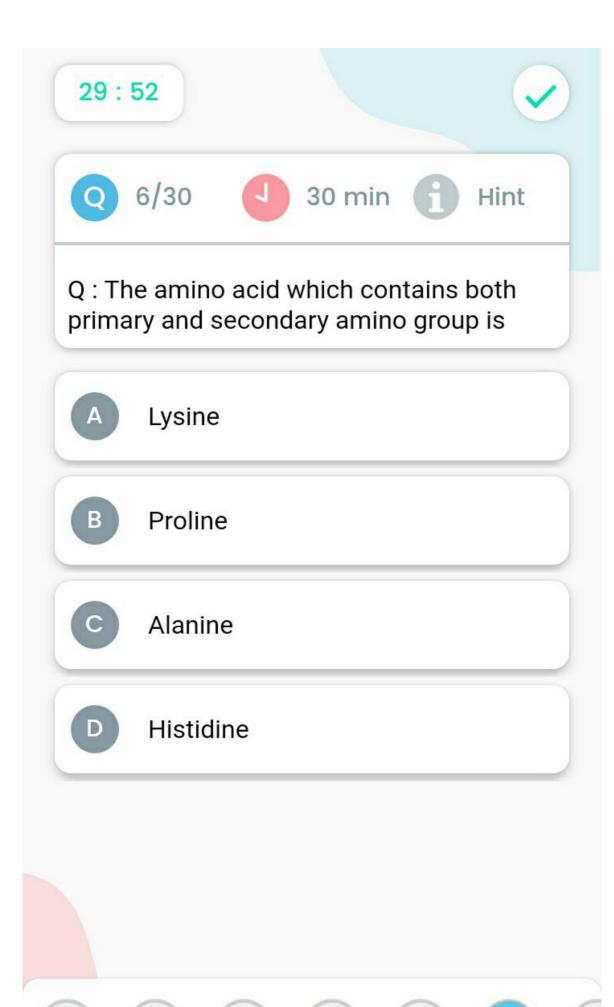




1 2 3 4 5 6 7







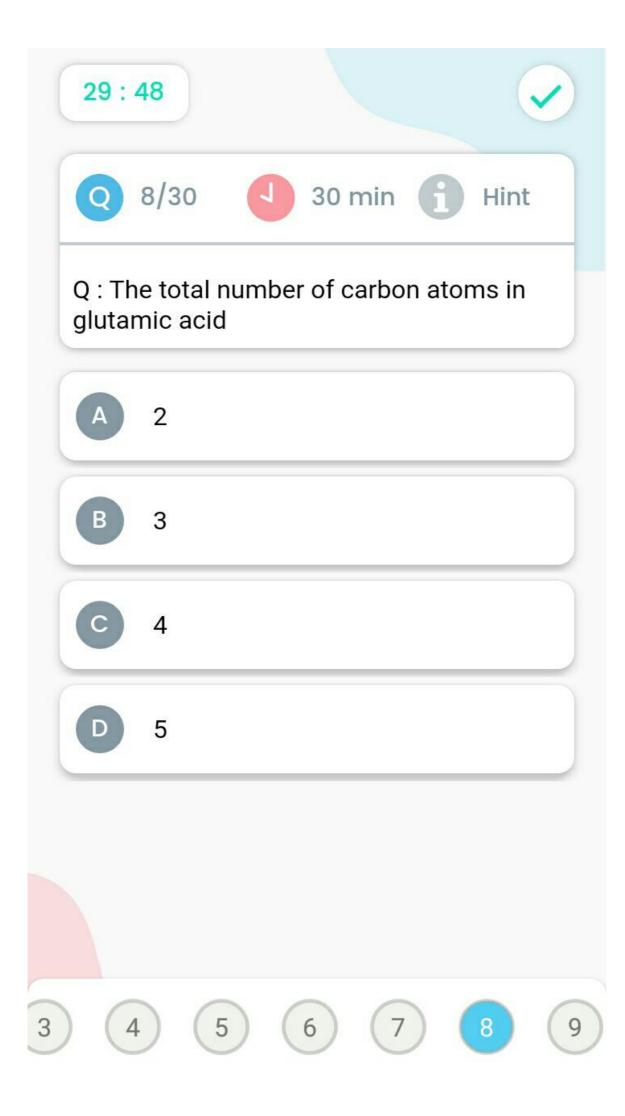
1 2 3 4 5 6

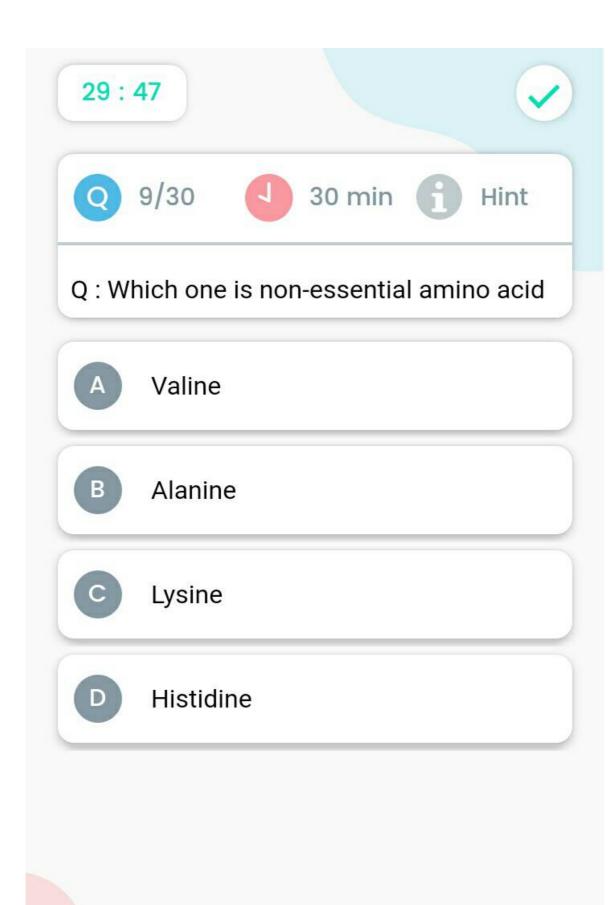


- 7/30
- 30 min Hint

Q : Name of given amino acid

- Alanine
- **Proline** В
- Valine
- Lysine





3) (4)

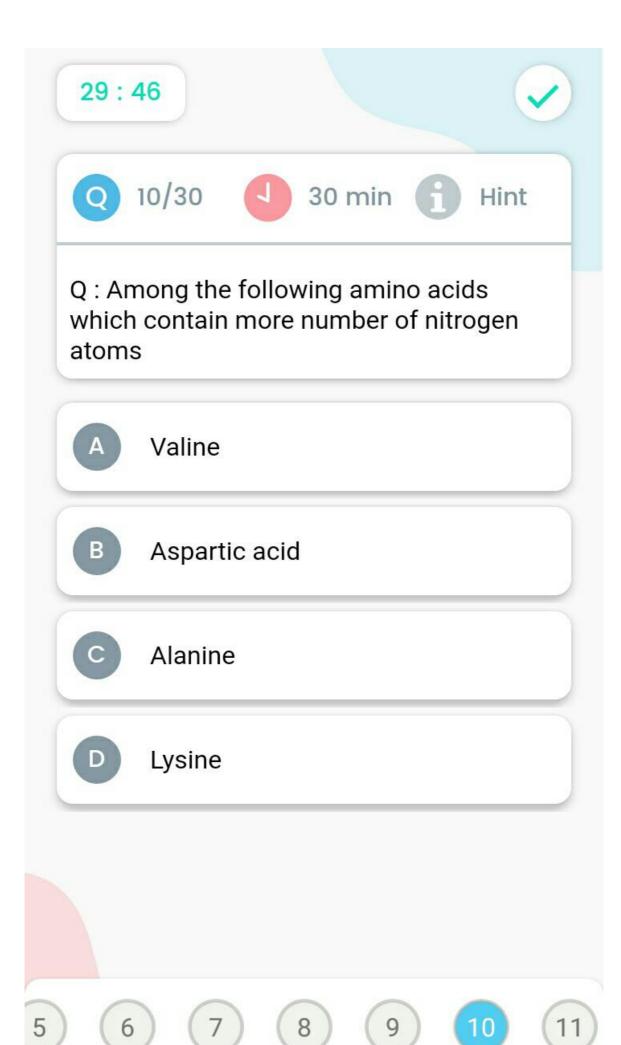


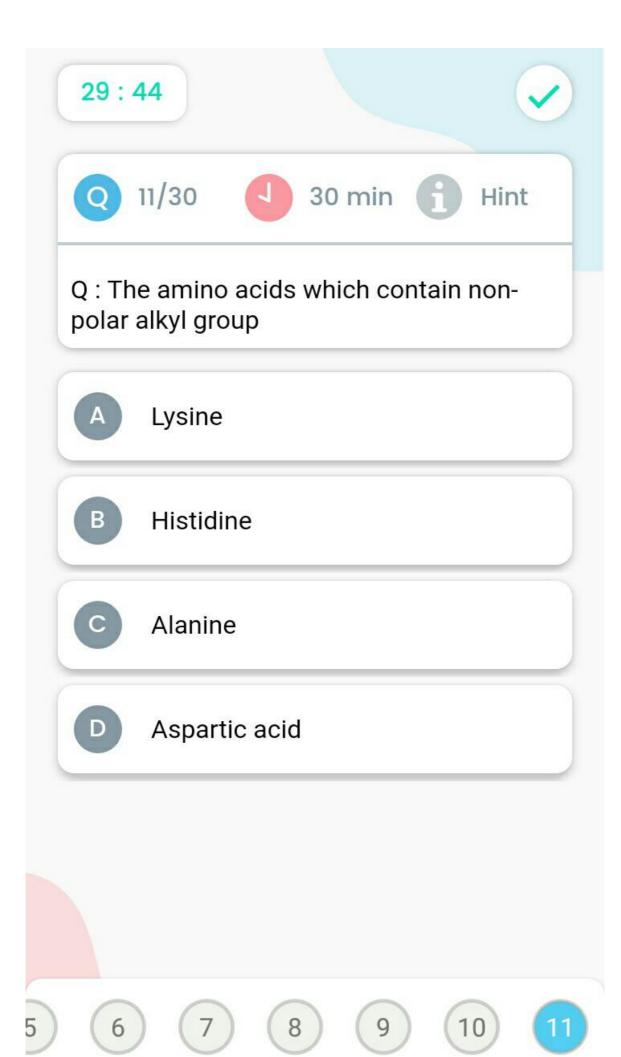


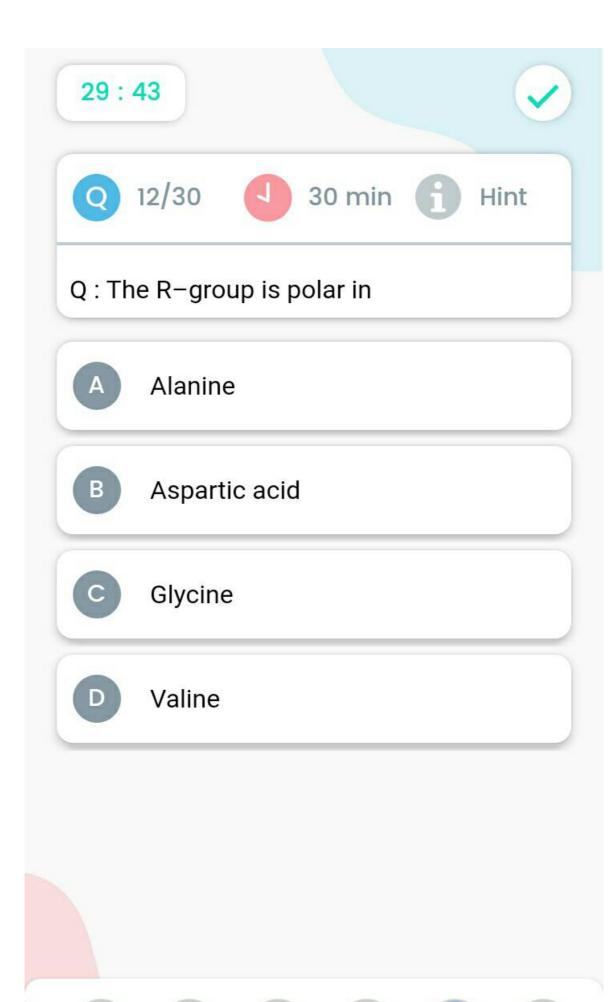










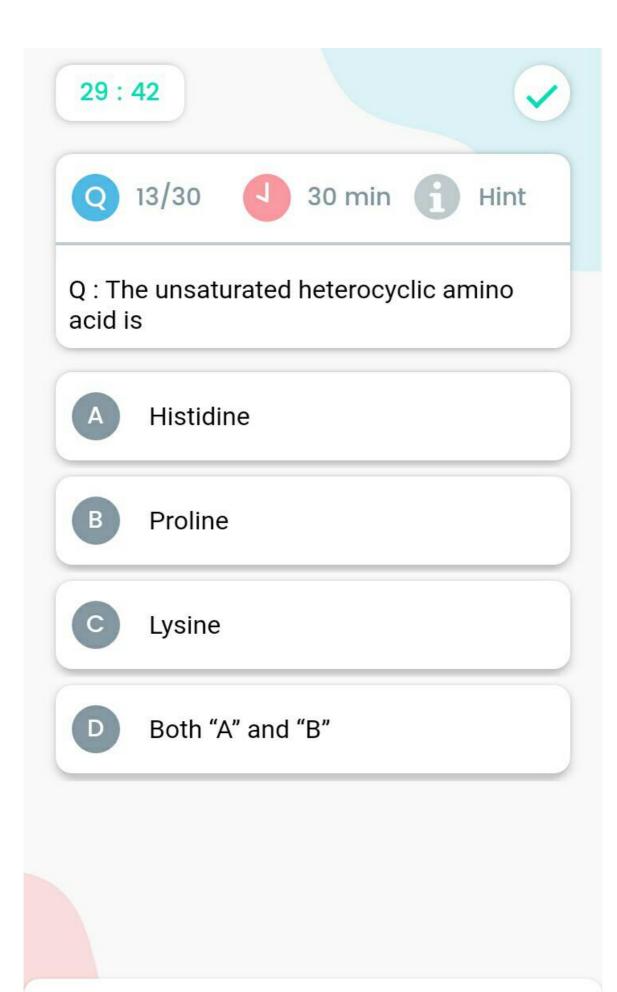
































Hint

Q: What is true about valine

- It must be supplied to our bodies through our diet
- B Its deficiency in our diet may not cause diseases
- It does not show optical isomerism
- It is a basic amino acid

9



11

















Hint

Q : The following carboxylic acid is used to synthesized alanine amino acid

- A Methanoic acid
- B Ethane dioic acid
- Propane dioic acid
- Propanoic acid

9

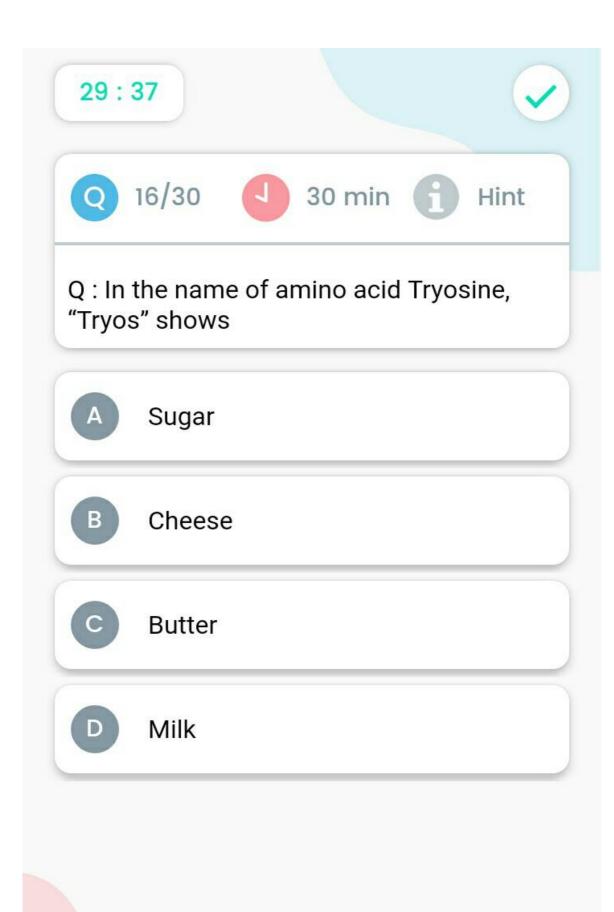
10

11

12

13

14









- 17/30

30 min 🚹 Hint



Q: Which one is not amino acid

- Aspartic acid
- Picric acid
- Glutamic acid
- Valine









30 min 👔



Hint

Q : There are three peptide bonds present in a molecule of a

- Tripeptide
- Tetrapeptide
- Pentapeptide
- Dipeptide







Q: Number of chiral centers in

2

В

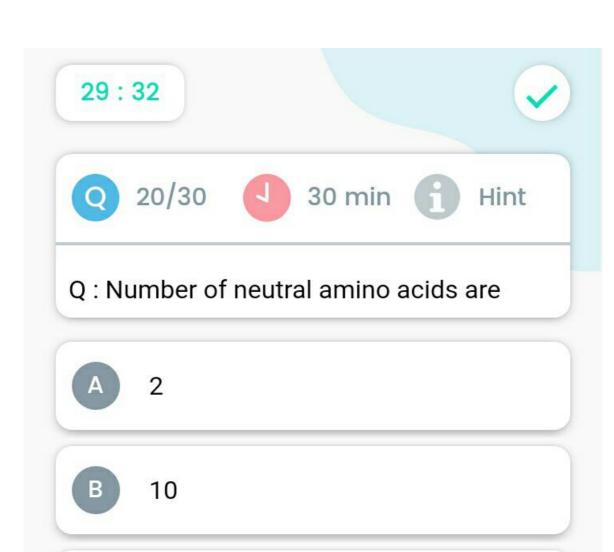
1



3

`4

(16)







4 15 16 17 18 19 20



- Q 21/30
- 30 min 🔒
 - Hint

Q: The name of the amino acid

H₂N -CH₂ -C - NH - CH - COOH CH₃

- A Alanylglycine
- B Glycylalanine
- C Lysylglycine
- Glycylvaline











Hint

Q: Which of the following amino acid has polar R group

- Glutamic acid
- Glycine
- Valine
- Alanine













Hint

Q: Which of the following is not an aliphatic amino acid?

- Glycine
- Valine
- Alanine
- Histidine





















Hint

Q: The acidic character of amino acids is due to

- Amino group
- Carbonyl group
- Carboxylate ion
- Carboxyl group









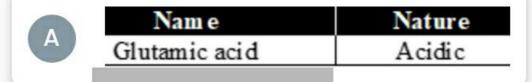
25/30 **3**0 min





Hint

Q: Which of the following pair is incorrect?



В	Nam e	Nature
	Glycine	Neutral

Name	Nature
Alanine	Acidic









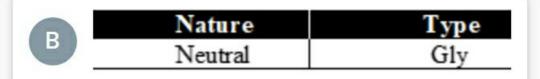


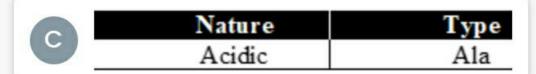


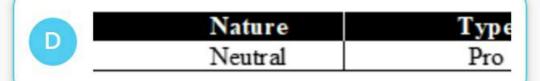
Hint

Q: Which of the following pair is incorrect?



















Q: Which of the following is non-essential amino acids?

- Lysine
- Valine
- Histidine
- **Proline**









Q: The structure of valine is

- CH2-COOH NH,
- CH₃-CH-COOH NH,
- CH₃-CH-CH-COOH CH, NH2
- H₂C-CH₂ H₂C CHCOOH

- 26











Hint

Q : Side chain in the structure of α -amino acid is represented by

СООН

R В

 NH_2

Н

26

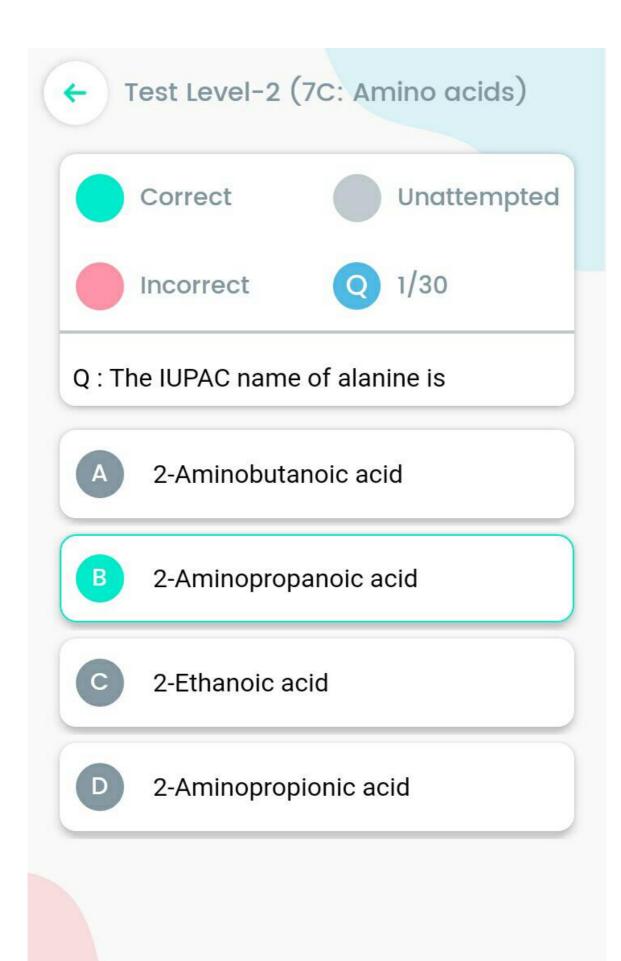


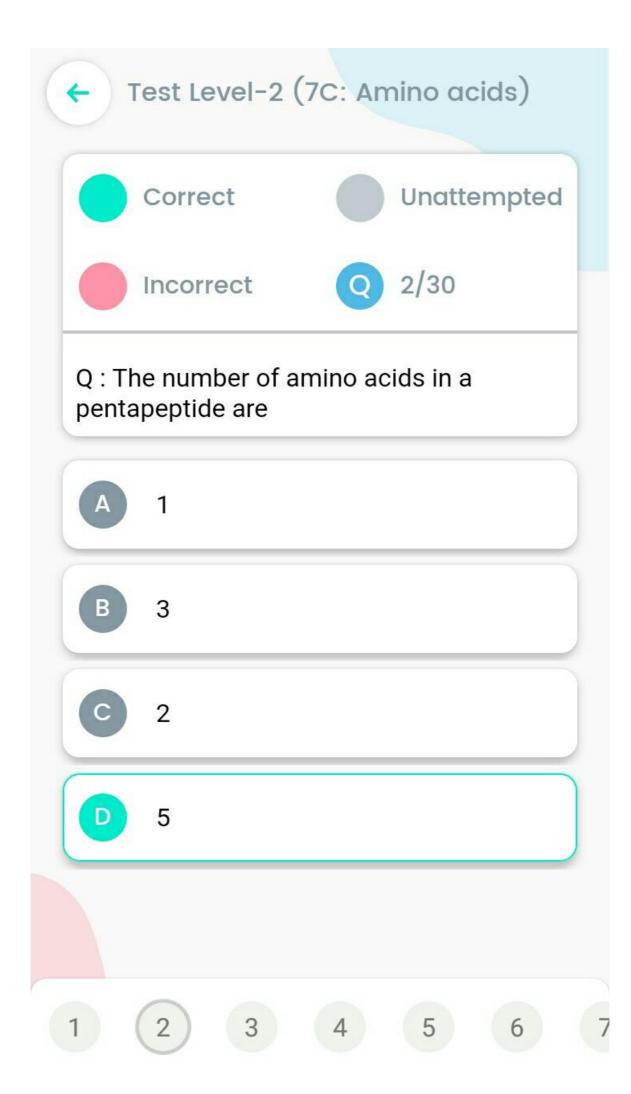
- 30/30

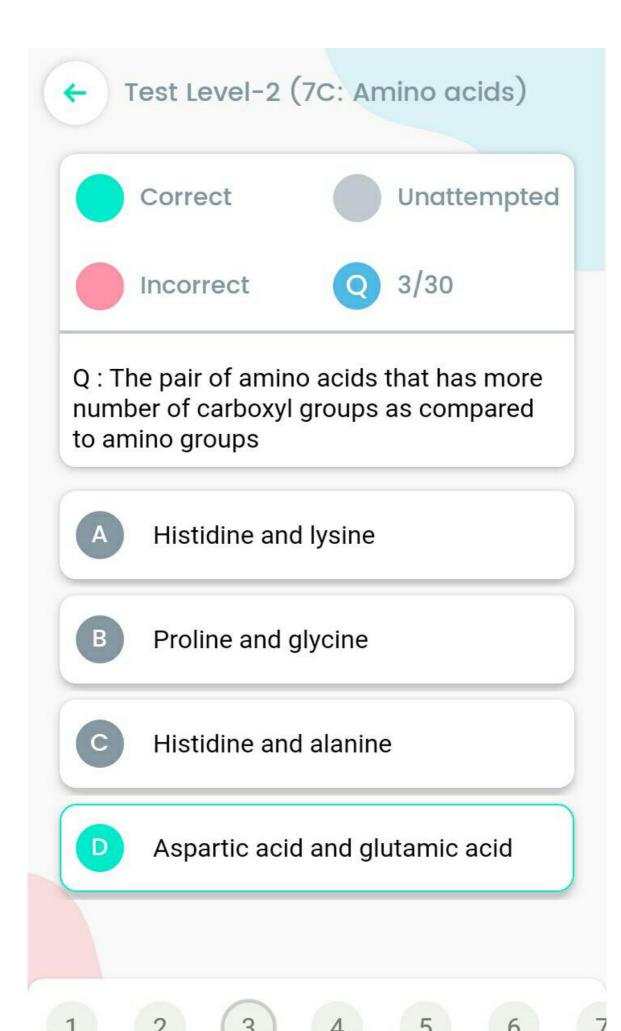
30 min Hint

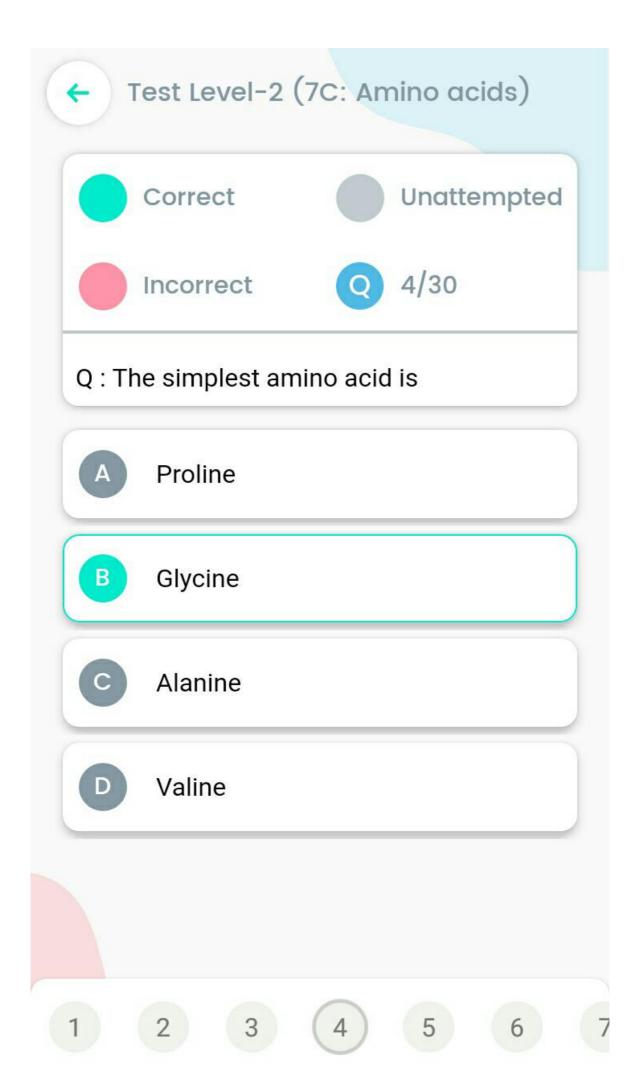
Q: a-carbon of nature occurring amino acid is always ____ hybridized

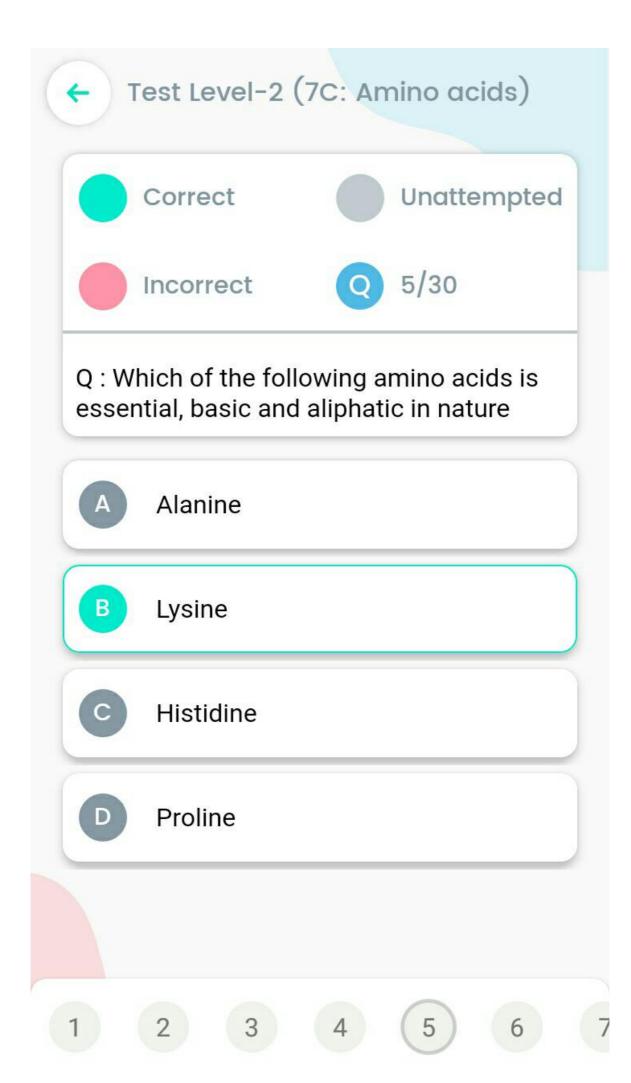
- sp
- sp²
- sp^3
- dsp^2

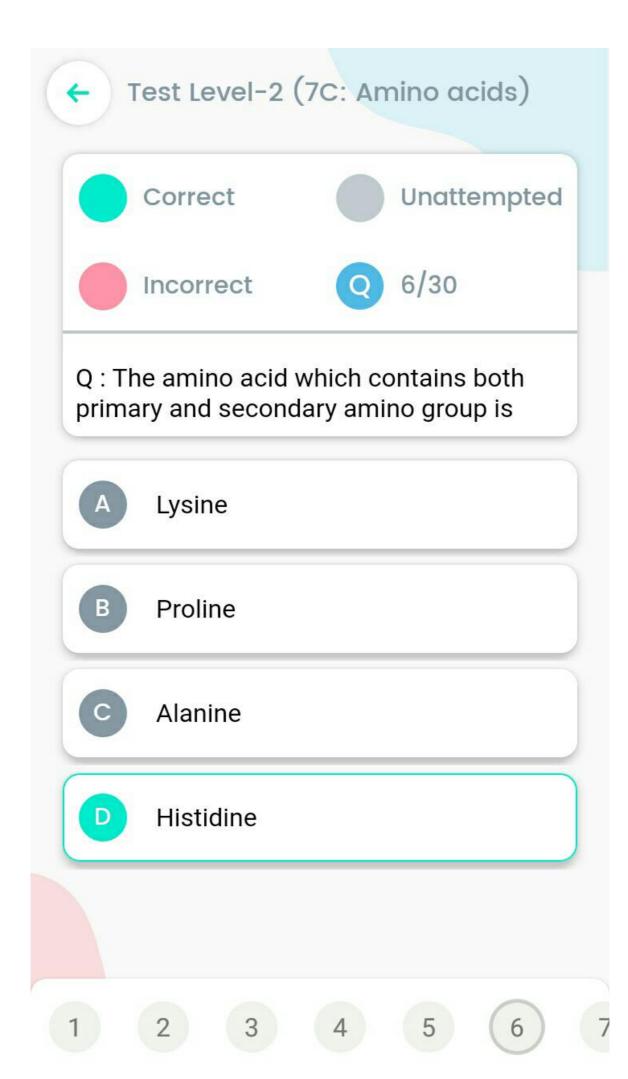










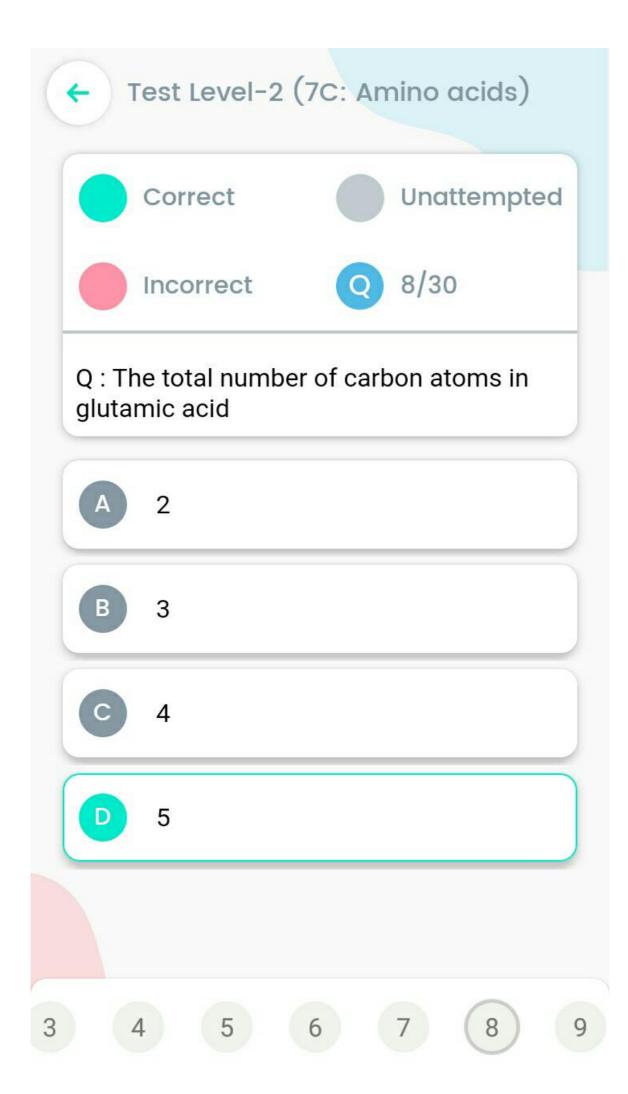


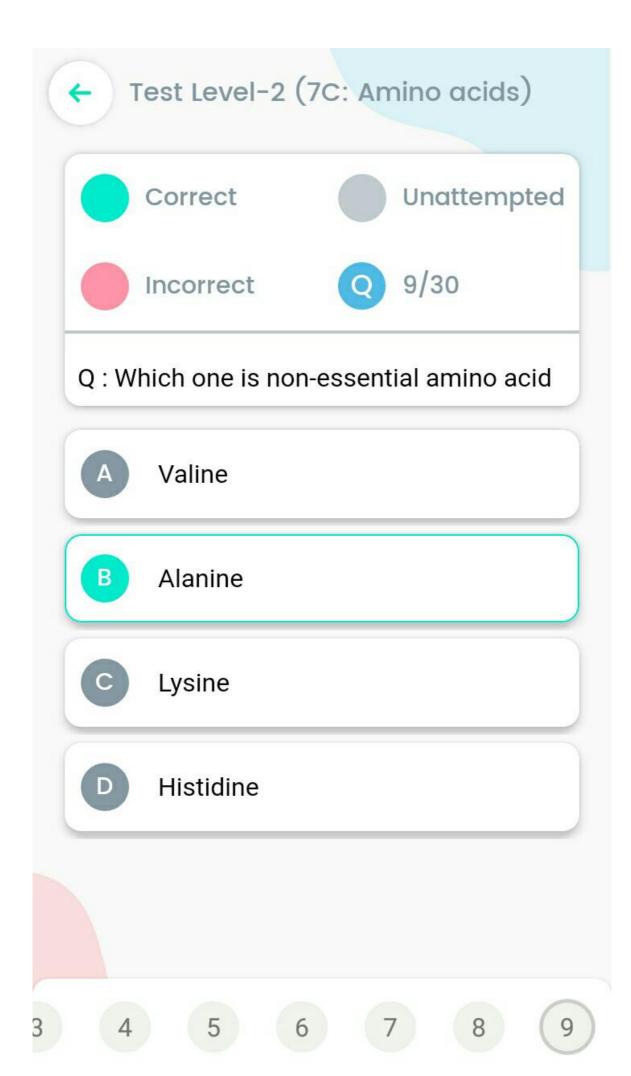


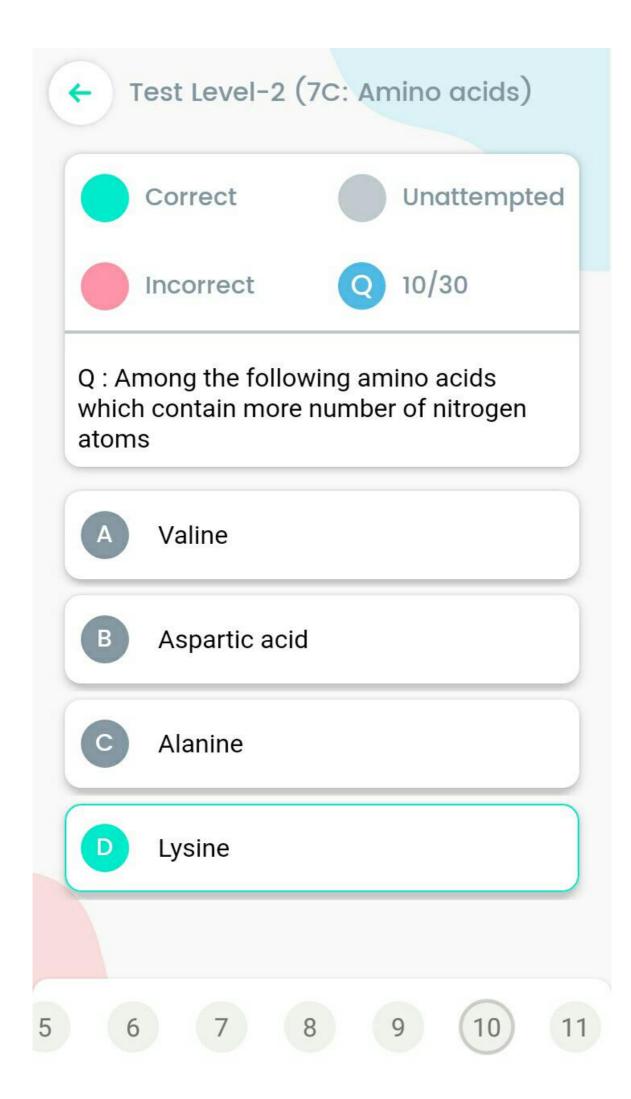
- Correct Unattempted
- Incorrect Q 7/30

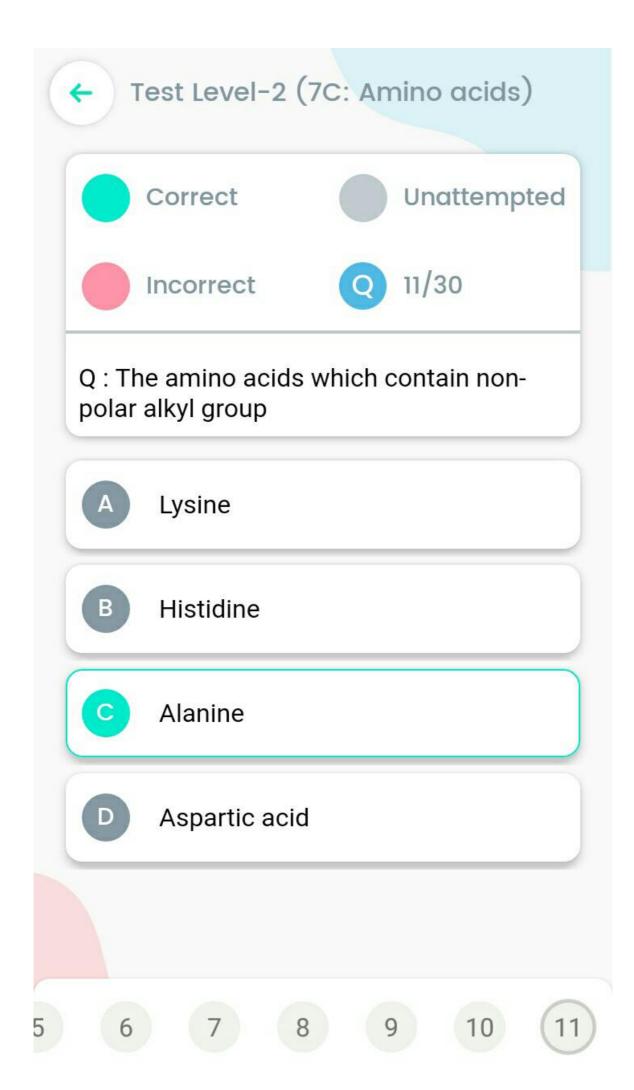
Q: Name of given amino acid

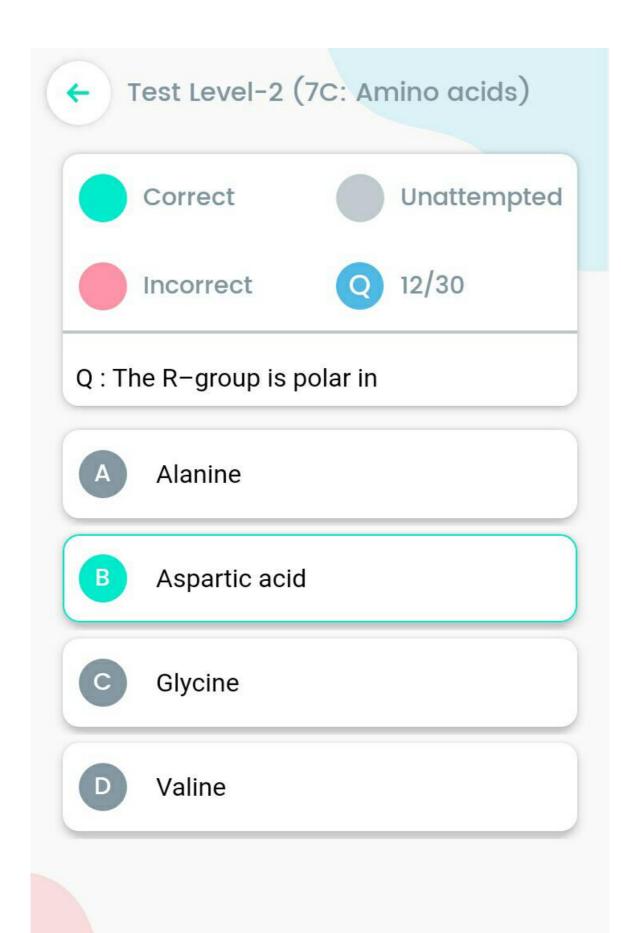
- A Alanine
- B Proline
- C Valine
- Lysine

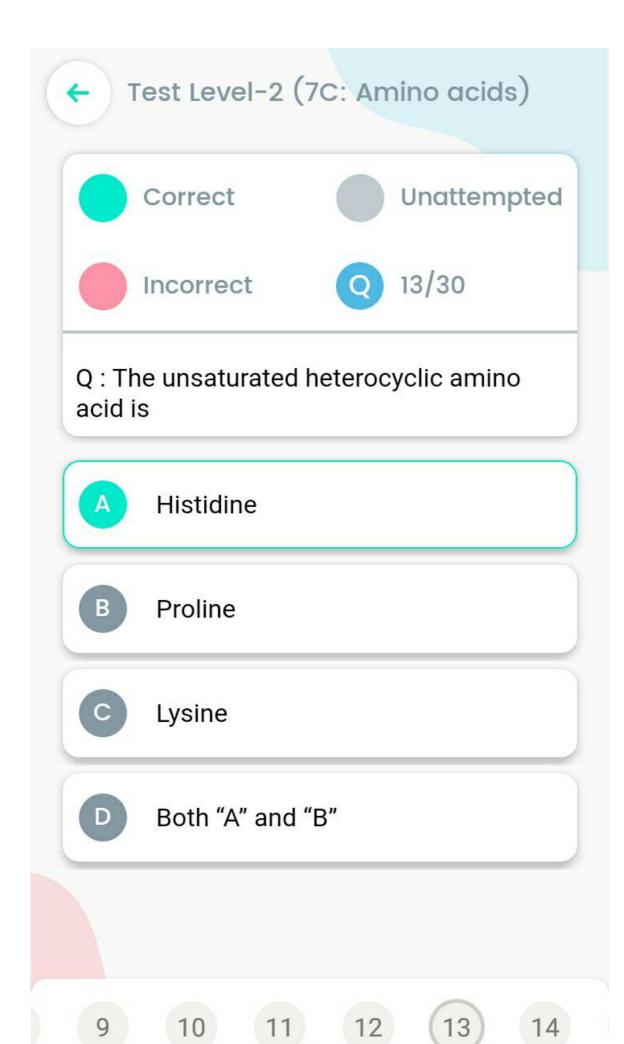










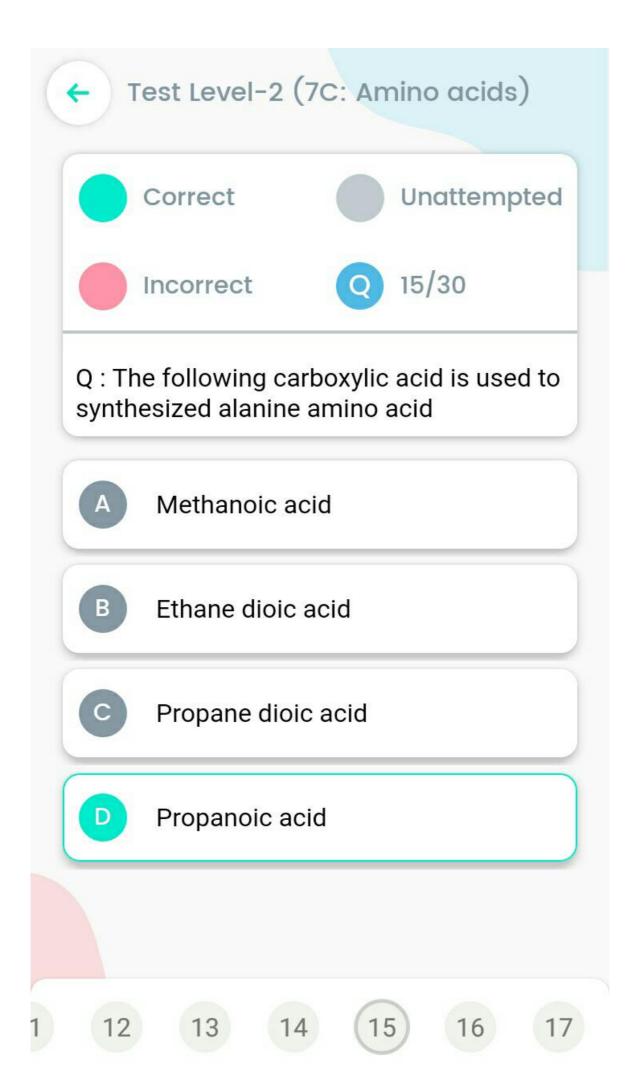


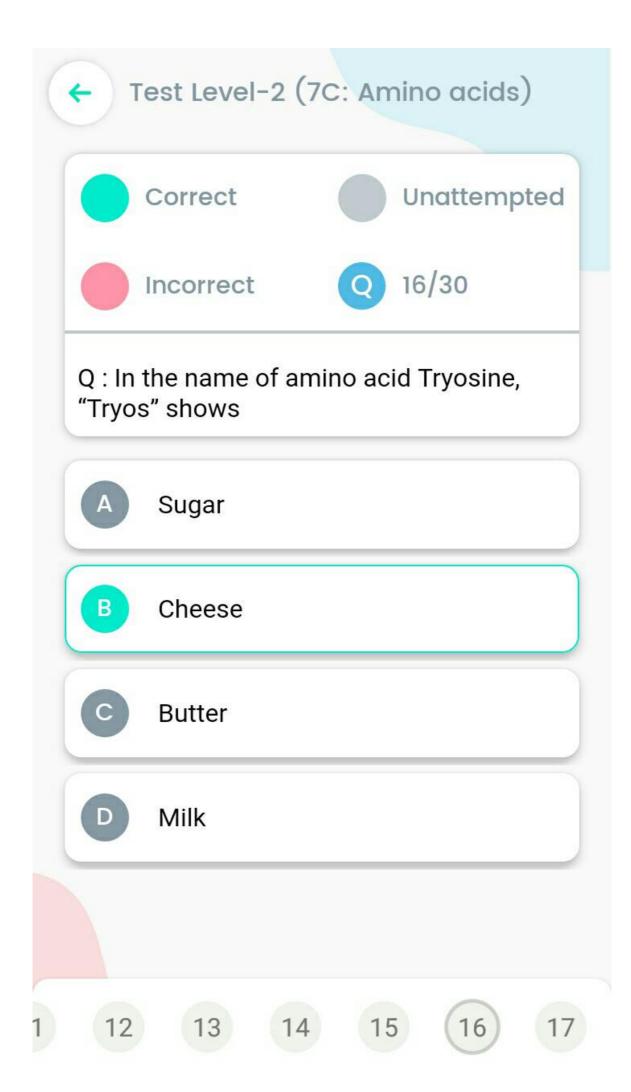


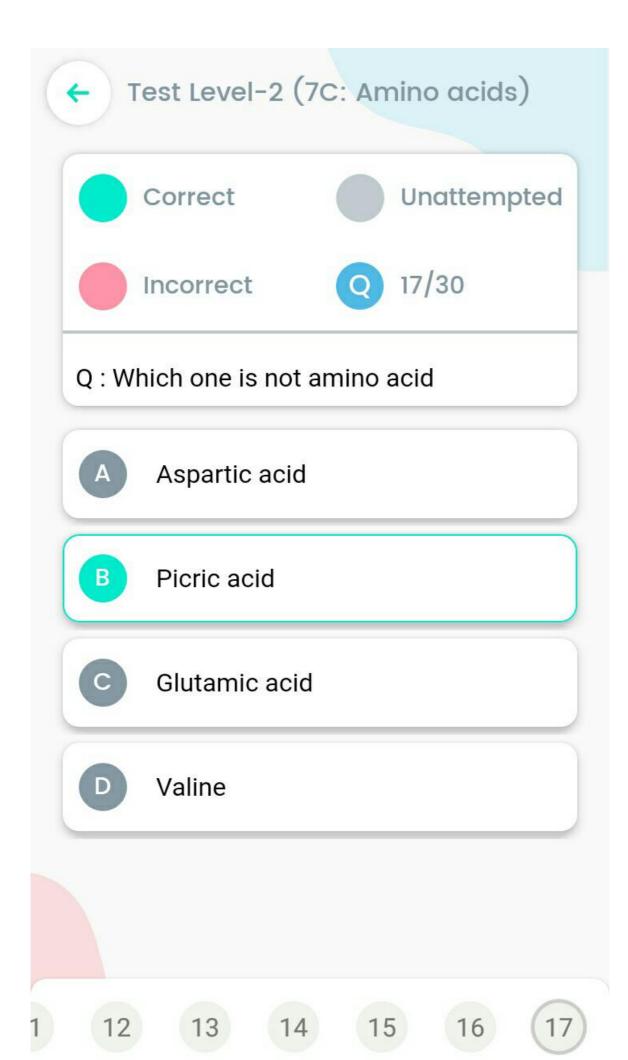
- Correct Unattempted
- Incorrect Q 14/30

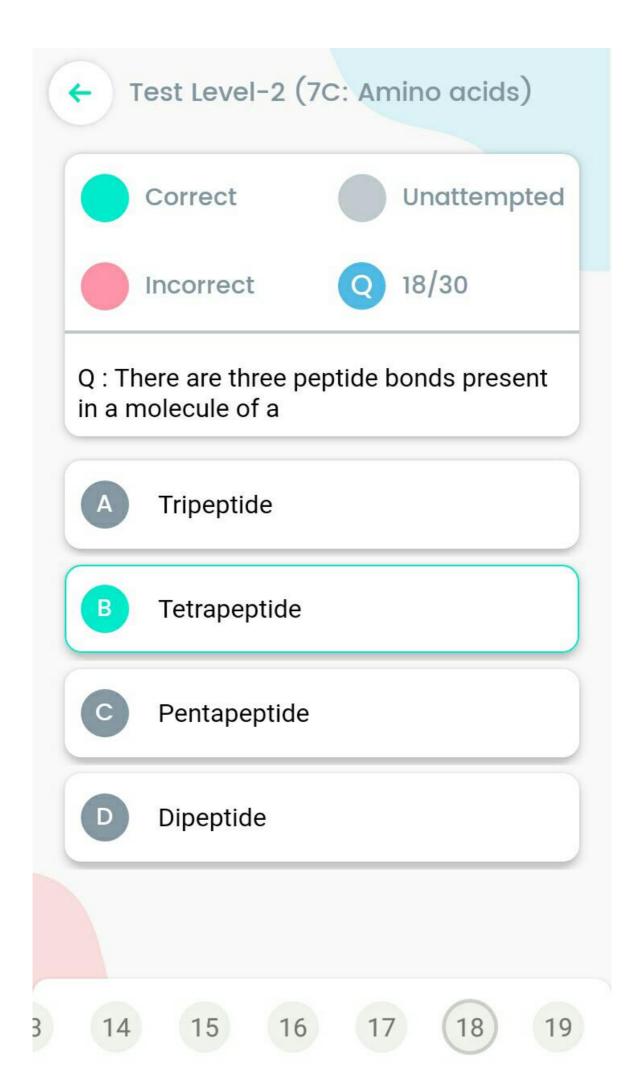
Q: What is true about valine

- It must be supplied to our bodies through our diet
- Its deficiency in our diet may not cause diseases
- It does not show optical isomerism
- It is a basic amino acid















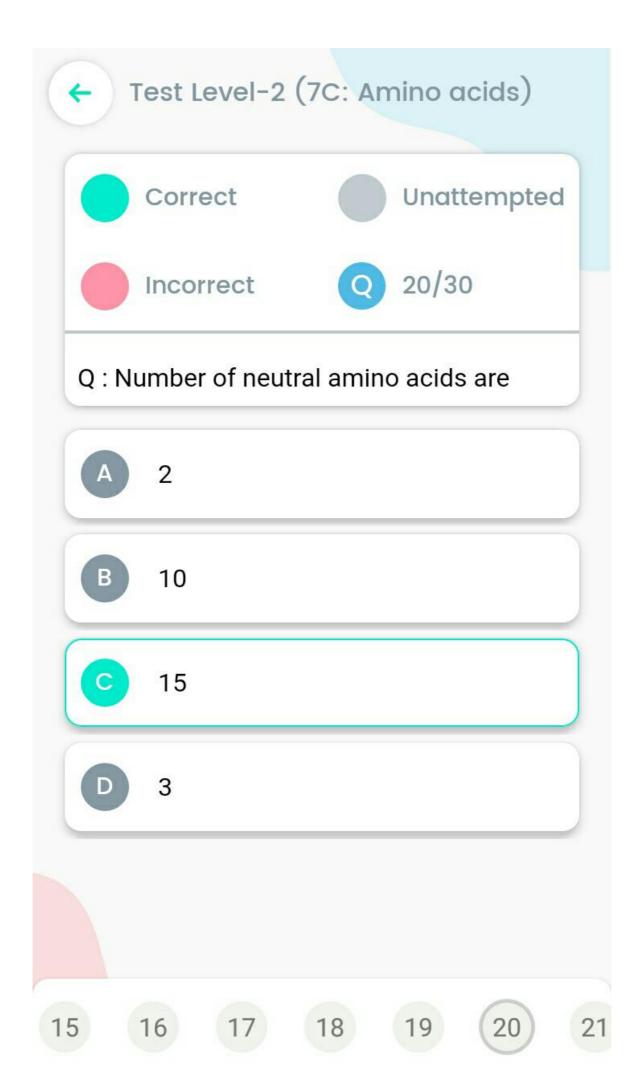
Q: Number of chiral centers in

A 2

B 1

C 3

D '4





- Correct Unattempted
- Incorrect Q 21/30

Q: The name of the amino acid

A

В

C

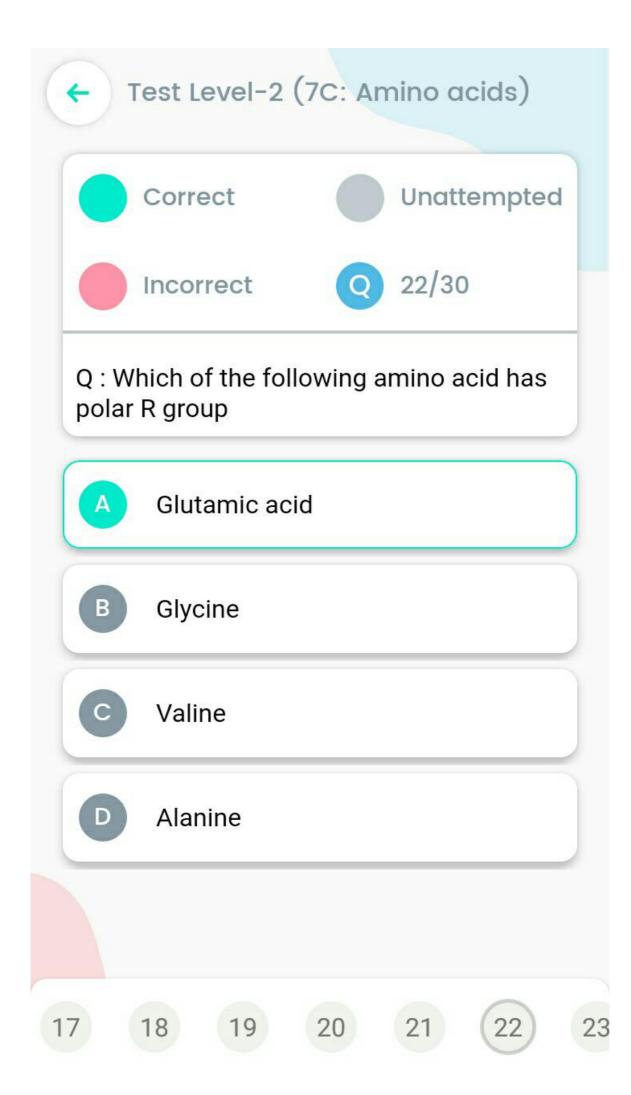
D

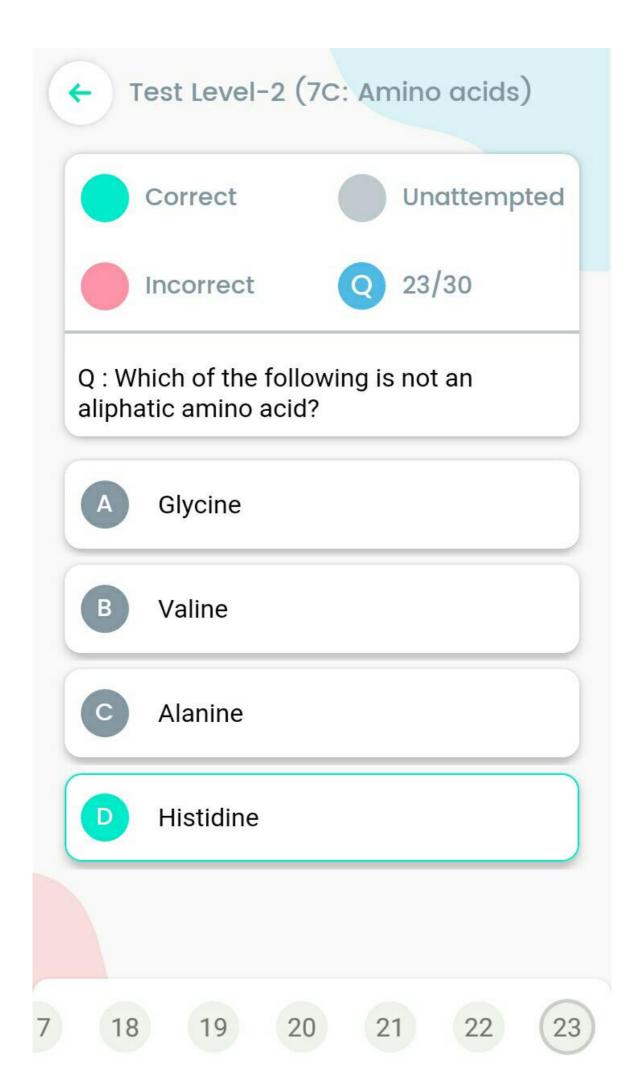


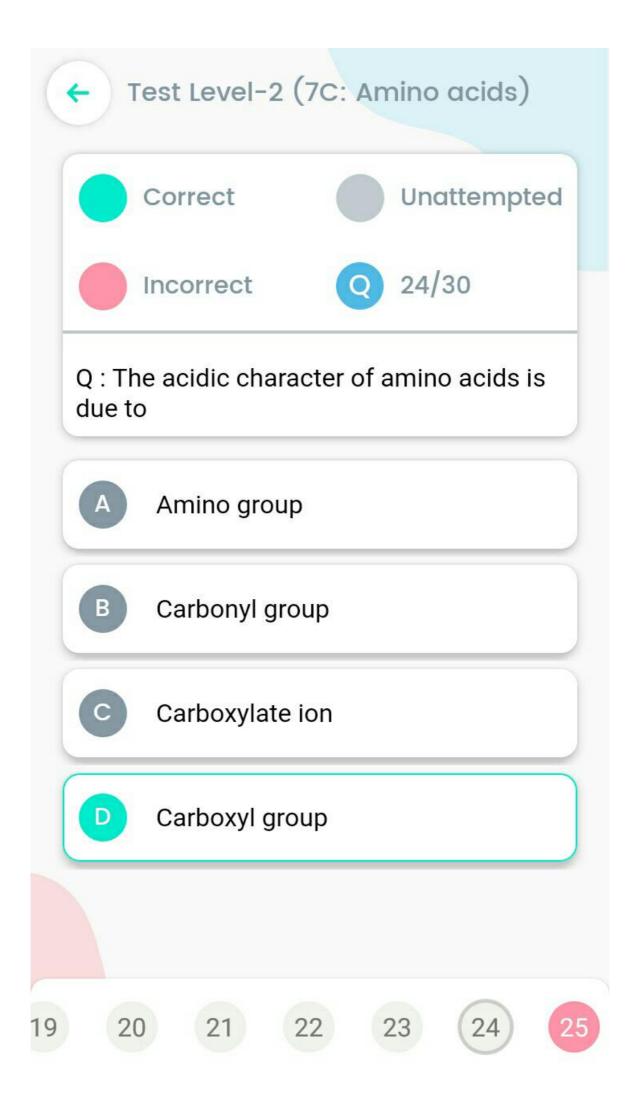
- Correct Unattempted
- Incorrect Q 21/30

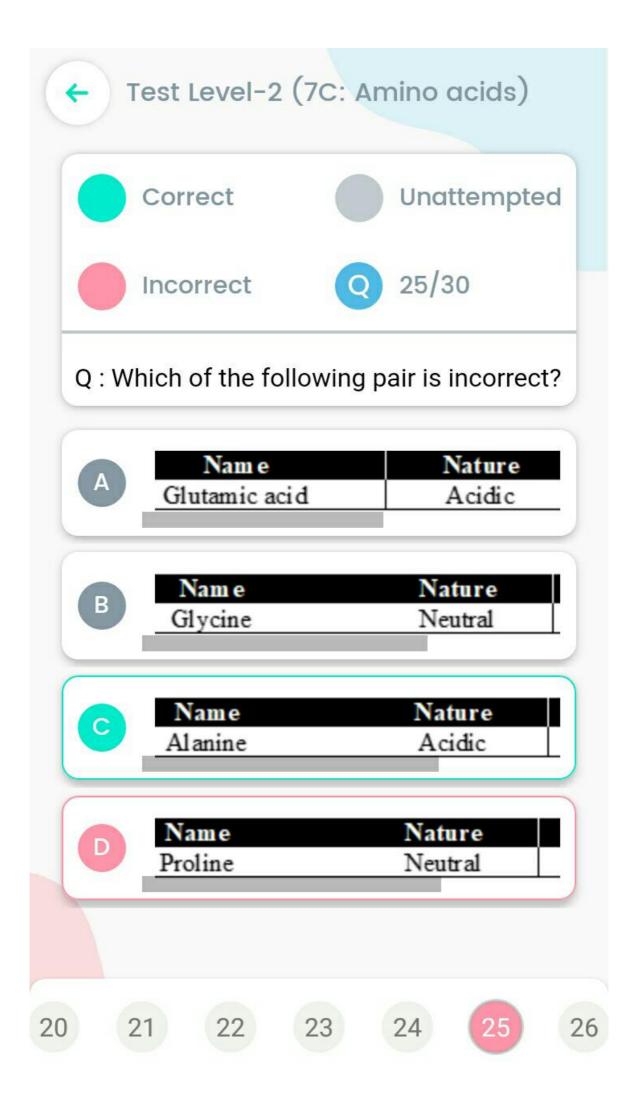
Q : The name of the amino acid

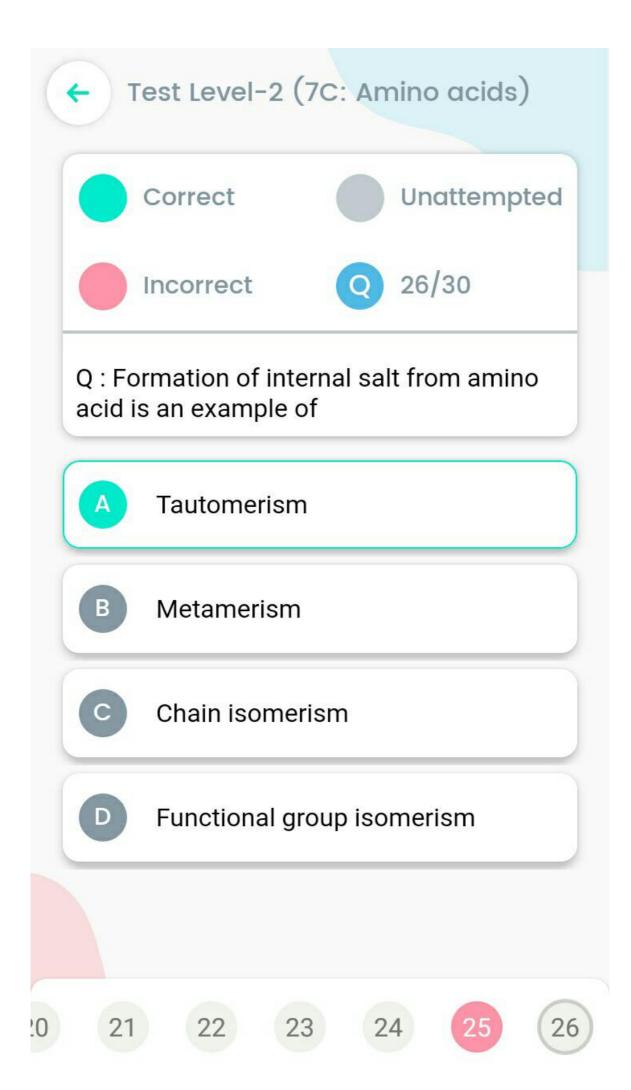
- A Alanylglycine
- B Glycylalanine
- Lysylglycine
- Glycylvaline

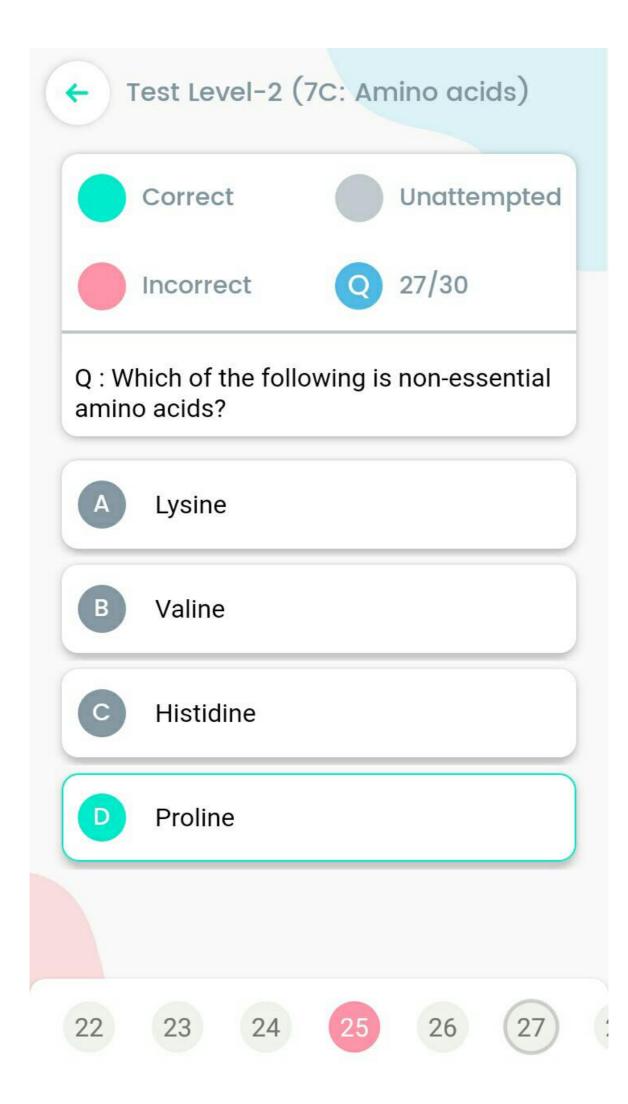














- Correct Unattempted
- Incorrect Q 28/30

Q: The structure of valine is

H₂C—СН₂ | | | | H₂С СНСООН

