

**Founders :**  
**Muhammad Kamran**  
**Fizza Marium**

**Motto :**  
**“We are saviour of nation.”**



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- 1) Wood, cotton and paper contain:  
 A) Starches  
 B) Chitin  
 C) Cellulose  
 D) Pectin
- A     B     C     D

- 2) Carbohydrates are composed of:  
 A) Carbon, Nitrogen and Oxygen  
 B) Carbon, Hydrogen, Oxygen and Nitrogen  
 C) Carbon, Hydrogen, Oxygen and Phosphorus  
 D) Carbon, Hydrogen and Oxygen
- A     B     C     D

- 3) The example of polyhydroxy ketone is:  
 A) Glucose  
 B) Glyceraldehyde  
 C) Dihydroxyacetone  
 D) Ribose
- A     B     C     D

- 4) In free state, glucose is abundant in:  
 A) All fruits  
 B) Grapes, Figs and Dates  
 C) Grapes, Guava and Dates  
 D) Grapes, Figs and Onion
- A     B     C     D

- 5) They are sweet in taste:  
 A) Proteins  
 B) Carbohydrates  
 C) Monosaccharides  
 D) Oligosaccharides
- A     B     C     D

- 6)  $C_3H_6O_3$  is the chemical formula of:  
 A) Glyceraldehyde  
 B) Dihydroxyacetone  
 C) Aldo-sugar  
 D) Triose
- A     B     C     D

- 7) Pick up the aldo-triose:  
 A) DAP  
 B) RUBP  
 C) G.6.  
 D) G.3.P
- A     B     C     D

- 8) The ketonic form of glyceraldehyde is:  
 A) Glucose  
 B) Dihydroxyacetone  
 C) Phosphoglycerate  
 D) Erythrose
- A     B     C     D

- 9) Monosaccharides may be:  
 A) Trioses to tetroses  
 B) Trioses to pentoses  
 C) Trioses to hexoses  
 D) Trioses to heptoses
- A     B     C     D

- 10) \_\_\_\_\_ and \_\_\_\_\_ are most common monosaccharides:  
 A) Trioses, tetroses  
 B) Tetroses, pentoses  
 C) Pentoses, hexoses  
 D) Hexoses, heptoses
- A     B     C     D

- 11) Ribose forms a:  
 A) Three cornered ring  
 B) Four cornered ring  
 C) Six cornered ring  
 D) Five cornered ring
- A     B     C     D

- 12) \_\_\_\_\_ corners of fructofuranose are occupied by carbon atoms:  
 A) Three  
 B) Six  
 C) Four  
 D) Five
- A     B     C     D

- 13) In free state, glucose is abundant in:  
 A) All fruits  
 B) Grapes, Figs and Dates  
 C) Grapes, Guava and Dates  
 D) Grapes, Figs and Onion
- A     B     C     D

- 14) Starch, cellulose and glycogen yield \_\_\_\_\_ on complete hydrolysis:  
 A) Fructose  
 B) Mannose  
 C) Glucose  
 D) Galactose
- A     B     C     D

- 15) Glucose is \_\_\_\_\_ in the body, to make the chemical energy available to organism:  
 A) Reduced  
 B) Oxidized  
 C) Hydrolyzed  
 D) Hydrated
- A     B     C     D

- 16) Two glucose molecules are obtained by the hydrolysis of:  
 A) Maltose  
 B) Raffinose  
 C) Sucrose  
 D) Lactose
- A     B     C     D

- 17) Glucose and fructose chemically combine to form sucrose by forming:  
 A) 1,4 glycosidic bond  
 B) 1,6 glycosidic bond  
 C) 1,2 glycosidic bond  
 D) 1,3 glycosidic bond
- A     B     C     D

- 18) How many carbon atoms are kept outside the ring in fructose?  
 A) 01  
 B) 02  
 C) 03  
 D) 04
- A     B     C     D

- 19) \_\_\_\_\_ are usually branched and tasteless:  
 A) Monosaccharides  
 B) Oligosaccharides  
 C) Polysaccharides  
 D) Disaccharides
- A     B     C     D

- 20) The main source of carbohydrates for animals is:  
 A) Glycogen  
 B) Starch  
 C) Cellulose  
 D) Dextrin
- A     B     C     D

- 21) These have branched chains and are insoluble in hot or cold water:  
A) Amylose  
B) Glycogen  
C) Cellulose  
D) Amylopectin
- A    B    C    D
- 22) Animal starch is the name used for:  
A) Glycogen  
B) Amylose  
C) Amylopectin  
D) Cellulose
- A    B    C    D
- 23) \_\_\_\_\_ gives red color with iodine:  
A) Starch  
B) Glycogen  
C) Amylose  
D) Amylopectin
- A    B    C    D
- 24) Which one of the following is relatively stable polysaccharide?  
A) Glycogen  
B) Amylose  
C) Cellulose  
D) Agar
- A    B    C    D
- 25) It gives no color with iodine:  
A) Amylose  
B) Amylopectin  
C) Glycogen  
D) Cellulose
- A    B    C    D
- 26) Which one of the following is a heterogeneous groups of organic biomolecules?  
A) Carbohydrates  
B) Lipids  
C) Proteins  
D) Nucleic acids
- A    B    C    D
- 27) Hydrophobicity is the characteristic of:  
A) Carbohydrates  
B) Proteins  
C) Lipids  
D) Nucleic acids
- A    B    C    D
- 28) Lipids store \_\_\_\_\_ amount of energy, as compared to the equal amount of carbohydrates:  
A) Double  
B) Triple  
C) Quadruple  
D) Quintuple
- A    B    C    D
- 29) Living bodies get insulation with the help of:  
A) Carbohydrates  
B) Lipids  
C) Nucleic acids  
D) Proteins
- A    B    C    D
- 30) Pick up the one which is not a lipid:  
A) Acyl glycerol  
B) Waxes  
C) Chitin  
D) Steroid
- A    B    C    D

- 31) Widespread acylglycerols are:  
 A) Tetraglycerides                      C) Diglycerides  
 B) Triglycerides                          D) Tetraglycerols
- A     B     C     D

- 32) Chemically \_\_\_\_\_ can be defined as esters of fatty acids and alcohol:  
 A) Phospholipids                      C) Acylglycerols  
 B) Waxes                                  D) Terpenoids
- A     B     C     D

- 33)  $C_2H_5OH + HOOCCH_3 \rightarrow C_2H_5OCOCH_3 + H_2O$   
 (alcohol)+(acetic acid)              (?) + (water)  $\rightarrow$   
 Name the product with “?” mark:  
 A) Fatty acid                              C) Ester  
 B) Amino acid                            D) Zwitterion
- A     B     C     D

- 34) During the formation of triacylglycerol, we get:  
 A) One water molecules              C) Three water molecules  
 B) Two water molecules              D) Four water molecules
- A     B     C     D

- 35) Pick up the one that always have even number of carbon atoms:  
 A) Proteins                                C) Carbohydrates  
 B) Lipids                                    D) Fatty acids
- A     B     C     D

- 36) Fatty acids may contain:  
 A) No double bond                      C) Upto 6 double bonds  
 B) One double bond                    D) No or upto six double bonds
- A     B     C     D

- 37) In plants the fatty acids are:  
 A) Branched or straight chain  
 B) Straight chain or ringed  
 C) Ringed or cyclic  
 D) Branched or ringed
- A     B     C     D

- 38) \_\_\_\_\_ is much more soluble in organic solvent than \_\_\_\_\_:  
 A) Acetic acid, butyric acid  
 B) Butyric acid, palmitic acid  
 C) Palmitic acid, butyric acid  
 D) Palmitic acid, oleic acid
- A     B     C     D

- 39) Melting point of butyric acid is:  
 A) 61.3 °C                                  C) 80 °C  
 B) 63.1 °C                                  D) -8 °C
- A     B     C     D

- 39) Melting point of butyric acid is:  
A) 61.3 °C                      C) 80 °C  
B) 63.1 °C                      D) -8 °C  
 A     B     C     D
- 40) A saturated fatty acid with two carbon atoms is:  
A) Oleic acid                      C) Butyric acid  
B) Palmitic acid                  D) Acetic acid  
 A     B     C     D
- 41) A saturated fatty acid with sixteen carbon atoms is:  
A) Acetic acid                      C) Palmitic acid  
B) Butyric acid                      D) Oleic acid  
 A     B     C     D
- 42) In oleic acid there is a single double bond between carbon number \_\_\_\_ and \_\_\_\_:  
A) 9, 10                              C) 7, 8  
B) 8, 9                                D) 6, 7  
 A     B     C     D
- 43) Fat containing \_\_\_\_\_ fatty acids are usually liquid at room temperature and are said to be oils:  
A) Saturated                        C) Unsaturated  
B) Straight chain                  D) Saturated unbranched  
 A     B     C     D
- 44) \_\_\_\_\_ fats are solid at room temperature:  
A) Animal                            C) All  
B) Plant                                D) No  
 A     B     C     D
- 45) Fat and oils are:  
A) Lighter than water              C) Less viscous than water  
B) Havier than water                D) More denser than water  
 A     B     C     D
- 46) They are not crystalline but some can be crystallized under specific conditions:  
A) Phospholipids and waxes      C) Fats and oils  
B) Waxes and terpenoids          D) Phospholipids and terpenoids  
 A     B     C     D
- 47) Glycerol, fatty acids and phosphoric acid gives rise to:  
A) Phospholipid                      C) Phosphatidyl choline  
B) Phosphatidic acid                D) Phosphatidyl ethndine  
 A     B     C     D
- 48) A phosphatidic acid upon combining with a choline gives rise to:  
A) Sphingolipids                    C) Phospholipids  
B) Neutral lipids                    D) Fatty acids  
 A     B     C     D
- 49) The lipids frequently associated with cellular membranes are:  
A) Neutral lipids                      C) Phospholipids  
B) Triglycerides                      D) Acylglycerols  
 A     B     C     D
- 50) In phosphatidic acid fatty acids are attached to carbon no. \_\_\_\_\_ of glycerol:  
A) 01                                      C) 01 and 02 both  
B) 02                                      D) 03  
 A     B     C     D

- 51) The word carbohydrate literally means:  
A) Dehydrated carbons      C) Hydrated carbons  
B) Hydrolyzed carbons      D) Reduced carbons  
 A     B     C     D
- 52) Polyhydroxy aldehydes or ketones are:  
A) Carbohydrates      C) Lipids  
B) Proteins      D) Nucleic acids  
 A     B     C     D
- 53) Both glycoproteins and glycolipids are components of:  
A) Cell membrane      C) Egg membrane  
B) Plasma membrane      D) Biological membrane  
 A     B     C     D
- 54) Simple sugars are called:  
A) Polysaccharides      C) Disaccharides  
B) Oligosaccharides      D) Monosaccharides  
 A     B     C     D
- 55) The carbohydrates which are readily soluble in water are included in:  
A) Carbohydrates      C) Oligosaccharides  
B) Monosaccharides      D) Polysaccharides  
 A     B     C     D
- 56) Aldo-triose differs from keto-triose with respect to:  
A) Number of carbon atoms  
B) Position of carbon atoms  
C) Position of hydrogen and oxygen atoms  
D) Number of hydrogen and oxygen atoms  
 A     B     C     D
- 57) Pick up the keto-triose:  
A) Dihydroxyacetone      C) Glucose 6 phosphate  
B) Ribulose bisphosphate      D) Glyceraldehyde 3 phosphate  
 A     B     C     D
- 58) Carbohydrates with three to seven carbon atoms are called:  
A) Monosaccharides      C) Oligosaccharides  
B) Disaccharides      D) Polysaccharides  
 A     B     C     D
- 59) Tetroses are rare in nature and occur in some:  
A) Fungi      C) Bacteria  
B) Algae      D) Cyanobacteria  
 A     B     C     D
- 60) From biological point of view the most important hexose is:  
A) Glucose      C) Galactose  
B) Fructose      D) Mannose  
 A     B     C     D



61) **Five cornered rings are formed by \_\_\_\_\_, when in solution:**  
A) Ribose, glucose and ribulose  
B) Deoxyribose, glucose and fructose  
C) Fructose, ribose and galactose  
D) Ribose, fructose and deoxyribose

- A  B  C  D

62) **In free state, glucose is present in:**  
A) Grapes  
B) Figs  
C) Dates  
D) All fruits

- A  B  C  D

63) **Our blood normally contains:**  
A) 0.08% glucose  
B) 0.8% glucose  
C) 0.008% glucose  
D) 8.0% glucose

- A  B  C  D

64) **For the synthesis of 10 grams of glucose, the solar energy used is:**  
A) 7170.6 Kcal  
B) 7017.6 Kcal  
C) 717.6 Kcal  
D) 7107.6 Kcal

- A  B  C  D

65) **Most familiar disaccharide is:**  
A) Maltose  
B) Sucrose  
C) Lactose  
D) Cellobiose

- A  B  C  D

66) **On hydrolysis sucrose yields:**  
A) Disaccharides  
B) Oligosaccharides  
C) Non-reducing sugars  
D) Reducing sugars

- A  B  C  D

67) **Both glucose and fructose are six-carbon monosaccharides. In solution form both make following ring structures, respectively:**

- A) 6 cornered, 5 cornered  
B) 5 cornered, 6 cornered  
C) 6 cornered, 6 cornered  
D) 5 cornered, 5 cornered

- A  B  C  D

68) **Sucrose is formed by condensation of:**  
A) L-glucose and D-fructose  
B) D-glucose and D-fructose  
C) D-glucose and L-fructose  
D) L-glucose and L-fructose

- A  B  C  D

69) **They have high molecular weight and are only sparingly soluble in water:**

- A) Monosaccharides  
B) Disaccharides  
C) Oligosaccharides  
D) Polysaccharides

- A  B  C  D

70) **Starches are of \_\_\_\_\_ types:**  
A) One  
B) Two  
C) Three  
D) Four

- 71) Amylose starch is:  
 A) Soluble in cold water      C) Soluble in lukewarm water  
 B) Soluble in hot water      D) Insoluble in hot water
- A     B     C     D

- 72) Muscle and liver cells have much quantity of:  
 A) Starch                              C) Glycogen  
 B) Chitin                              D) Amylose
- A     B     C     D

- 73) Cotton is a pure form of:  
 A) Lipid                              C) Carbohydrate  
 B) Protein                            D) Phospholipid
- A     B     C     D

- 74) We are unable to digest:  
 A) Cellulose                          C) Amylose  
 B) Glycogen                          D) Amylopectin
- A     B     C     D

- 75) In the digestive tract of herbivores, cellulase is secreted by:  
 A) Bacteria, Fungi, Protozoa    C) Bacteria, Yeast, Protozoans  
 B) Bacteria, Yeast, Viruses      D) Bacteria, Fungi, viruses
- A     B     C     D

- 76) Insolubility in water and solubility in organic solvent is the characteristics of:  
 A) Nucleic acids                      C) Lipids  
 B) Carbohydrates                    D) Proteins
- A     B     C     D

- 77) Maximum energy storage is made by:  
 A) Lipids                              C) Nucleic acids  
 B) Proteins                            D) Carbohydrates
- A     B     C     D

- 78) Cuticle of plants is made up of:  
 A) Lipids                              C) Nucleic acids  
 B) Carbohydrates                    D) Proteins
- A     B     C     D

- 79) The class of lipids which is part of our daily life is named as:  
 A) Phospholipids                      C) Neutral lipids  
 B) Sphingolipids                      D) Glycolipids
- A     B     C     D

- 80) Acylglycerols are composed of:  
 A) Glycerol and fatty acids      C) Alcohol and fatty acids  
 B) Glycerol and amino acids      D) Alcohol and amino acids
- A     B     C     D

81) Following are the names of same class of lipids, EXCEPT:

- A) Triglycerides                      C) Sphingolipids  
B) Neutral lipids                      D) Acylglycerols

A     B     C     D

82) A compound produced as a result of a chemical reaction of an alcohol with an acid, where a water molecule is released, is called:

- A) Acylglycerol                      C) Fatty acids  
B) Ester                                  D) Wax

A     B     C     D

83) During the formation of a triglyceride \_\_\_\_\_ molecules of water are formed:

- A) 1    C) 3  
B) 2    D) 4

A     B     C     D

84) The number of carbon atoms in fatty acids may be:

- A) 2-7                                      C) 2-30  
B) 3-7                                      D) 2-40

A     B     C     D

85) Fatty acids contains even number of carbon atoms in straight chain attached with \_\_\_\_\_ and having \_\_\_\_\_:

- A) Hydrogen, an acidic group  
B) An acid, an hydrogen  
C) Hydrogen, an alcohol  
D) An alcohol, an acidic group

A     B     C     D

86) In animals the fatty acids are:

- A) Branched                              C) Ringed  
B) Straight chain                      D) Unsaturated

A     B     C     D

87) Solubility of fatty acids in organic solvents and their melting points increase with increasing the:

- A) Number of carbon atoms in chain  
B) Number of oxygen atoms in chain  
C) Number of hydrogen atoms in chain  
D) Number of acid group in chain

A     B     C     D

88) Melting point of palmitic acid is:

- A) 61.3° C                                  C) 80° C  
B) 63.1° C                                  D) -8° C

A     B     C     D

89) The smallest fatty acid is:

- A) Oleic acid                              C) Butyric acid  
B) Palmitic acid                          D) Acetic acid

A     B     C     D

90) **A saturated fatty acid with four carbon atoms is:**  
 A) Acetic acid C) Palmitic acid  
 B) Butyric acid D) Oleic acid

- A  B  C  D

91) **An unsaturated fatty acid with eighteen carbon atoms is:**  
 A) Acetic acid C) Palmitic acid  
 B) Butyric acid D) Oleic acid

- A  B  C  D

92) **For single double bond a fatty acid suffers from unsaturation of:**  
 A) One H-atoms C) Three H-atoms  
 B) Two H-atoms D) Four H-atoms

- A  B  C  D

93) **Fat containing unsaturated fatty acids are usually represented by:**

- A) Oils C) Butter  
 B) Solid fats D) Banaspati

- A  B  C  D

94) **\_\_\_\_\_ fats are solid at room temperature, whereas most of the \_\_\_\_\_ fats are liquids, respectively:**

- A) Plant, animal C) Synthetic, Natural  
 B) Animal, plant D) Natural, Synthetic

- A  B  C  D

95) **The specific gravity of fats and oils is:**

- A) 0.08 C) 0.008  
 B) 0.8 D) 8.0

- A  B  C  D

96) **Phospholipids are derivatives of:**

- A) Phosphatidic acid C) Lecithin  
 B) Phosphatidylcholine D) Phosphatidylserine

- A  B  C  D

97) **Phosphatidic acid upon combining with a nitrogenous base gives rise to:**

- A) Phosphatidylserine C) Phospholipids  
 B) Phosphatidylcholine D) Lecithin

- A  B  C  D

98) **Which of the following is/are frequently associated with membrane?**

- A) Phospholipids C) Lecithin  
 B) Phosphatidic acid D) Phosphatidylcholine

- A  B  C  D

99) **In phosphatidic acid, phosphoric acid is attached to carbon no. \_\_\_\_\_ of glycerol:**

- A) 01 C) 03  
 B) 02 D) 04

- A  B  C  D

100) **The non-polar region of a phospholipid molecule is:**

- A) Hydrophilic region  
 B) Phosphate containing region  
 C) Alcohol containing region  
 D) Fatty acid containing region

- A  B  C  D

# Medicos Hub Bio Test #2 Key

Key

1.	C	17.	C	33.	C	49.	C	65.	B	81.	C	97.	C
2.	D	18.	B	34.	C	50.	C	66.	D	82.	B	98.	A
3.	C	19.	C	35.	D	51.	C	67.	A	83.	C	99.	C
4.	B	20.	B	36.	D	52.	A	68.	C	84.	C	100.	D
5.	C	21.	D	37.	D	53.	D	69.	D	85.	A	101.	
6.	D	22.	A	38.	C	54.	D	70.	B	86.	B	102.	
7.	D	23.	B	39.	D	55.	B	71.	B	87.	A	103.	
8.	B	24.	C	40.	D	56.	C	72.	C	88.	B	104.	
9.	D	25.	D	41.	C	57.	A	73.	C	89.	D	105.	
10.	C	26.	B	42.	A	58.	A	74.	A	90.	B	106.	
11.	D	27.	C	43.	C	59.	C	75.	C	91.	D	107.	
12.	C	28.	A	44.	A	60.	A	76.	C	92.	B	108.	
13.	B	29.	B	45.	A	61.	D	77.	A	93.	A	109.	
14.	C	30.	C	46.	C	62.	D	78.	A	94.	B	110.	
15.	B	31.	B	47.	B	63.	A	79.	C	95.	B	111.	
16.	A	32.	C	48.	C	64.	C	80.	A	96.	A	112.	